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SDI FINAL EVALUATION FORM 1.1

PART 1:

Journal Name:	Advances in Research
Manuscript Number:	2014_AIR_9672
Title of the Manuscript:	Rapid chemical bath deposition and optical property of CuS films using sodium ethylenediamine tetraacetate as chelating agent

PART 2:

I believe this manuscript must not be published by any journal. I will not review this	
manuscript further.	
Final comments(FC) with the previous questions(Q) and answers(A):	
Q3) I cannot identify which are the particles in Figure 2. The authors should identify the	
particle by means of a circle or any type of indicators.	
A3). The particles are identified by indicators.	
FC3) I still cannot recognize which is the particle. Where is the boundary?	
Q4) The deposition rate at different EDTA-2Na:Cu ²⁺ conditions in Figure 3 should be added as	
the evidence which clearly show the existence of maximum deposition rate.	
A4). A plot of deposition vs. ED1A-2Na/Cu was inserted into the figure 3.	
FC4) I drew the following lighter using the plots at 1 nr in Figure 5, 1 added the line obtained using the least squares method. I believe the relationship between the thickness and EDTA 2Ne/Cu should	
the least squares method. I believe the relationship between the unickness and EDTA-2Na/Cu should be recognized like this if the date are limited to those only in this manuscript.	
be recognized like tills, if the data are finited to those only in this manuscript.	
e e e e e e e e e e e e e e e e e e e	
<u> 300</u>	
- $\frac{2}{2}$ 100 - Regression	
Thickness = -90 EDTA-2Na/Cu + 310	
0.0 0.5 1.0 1.5 2.0	
FDTA 2Na/Cu	
05) The composition of CuS and Cu ₂ S in the film should be given. The authors mentioned Cu ₂ S	
were covered with CuS in Figure 1. The considerable decrease in transmittance after long	
denosition, in Fig. 4 (b), was explained relating to higher transmittance of Cu_2S than CuS . This	
means that content of Cu_2S was maximum at EDTA-2Na: $Cu^{2+}=1.0$. Was the maximum	
deposition rate achieved by Cu ₂ S, not by CuS? If the contribution of Cu ₂ S was large at EDTA-	
$2Na: Cu^{2+}=1.0$, I must recognize that the authors chose the unsuitable condition for CuS	
deposition.	
A5). The samples were fabricated two years ago and now are lose, so the composition of CuS and	
CuS can not be given. Line 119121, the sentences related to higher transmittance of Cu ₂ S than	
CuS have been removed.	
FC5) Finally, the contradiction remained.	

Note: Anonymous Reviewer