

Simulation and Comparison of the performance of refrigerant fluids in single stage vapour compression refrigeration system

ABSTRACT

This paper presents reports on simulation and comparative analysis of single stage vapour compression refrigeration system performance for comparison of refrigerant fluids. A mathematical model was devised by applying the concept of energy balance in the thermodynamic cycle to the components of the vapour compression refrigeration system. The developed model implemented in **MATLAB[1]** software was used to compare the performance of the system using Hydrocarbons, R134a, and ammonia as refrigerants in place of refrigerants 12 and 22. Simulation data was generated over a wide range of evaporation and condensation temperatures of -25 to 15°C and 30 to 60°C respectively for the selected working fluids considered in this study to observe the performance of the system in terms of refrigerating effect, coefficient of performance (COP) and overall efficiency of the system. In the present study, close match of COP and the system overall efficiency values for, R134a, R290 and R600a and R290 are observed with that of R12 and R22 counterpart under the similar operating conditions. Moreover, among the refrigerants simulated in the system under the similar operating conditions, the system with the use of ammonia yields highest COPs and are of up to 6.3 and 7.5% increment compared with those of R12 and R22 respectively. It can be concluded that R134a, R290, R600a and ammonia may act as substitutes for R12 and R22 without affecting the energy efficiency of vapour compression refrigeration system, provided the safety issues are addressed in terms of manufacturing, handling, storage and servicing so as to prevent flammability and corrosive/toxic natures of both hydrocarbons and ammonia refrigerants respectively, thus helping to solve the ozone depletion potential (ODP) and global warming potential (GWP) problems regarding environmental issues.

Keywords: Vapour compression refrigeration system, refrigerant, coefficient of performance, refrigerating effect, overall efficiency of the system

1. INTRODUCTION

Vapour compression refrigeration system based applications make use of refrigerants such as chlorofluorocarbons (CFCs) and hydro chlorofluorocarbons (HCFCs) which are responsible for greenhouse gases, global warming and ozone layer depletion effects. These effects cause a lot of ill health and diseases for living beings. These refrigerants have been used widely over the last eight decades in refrigeration and air-conditioning systems due to their favourable characteristics such as low freezing point, non-flammability, non-toxicity and chemically stable behaviour with other materials, Mohan [2]. The role of these refrigerants in the process of ozone depletion is now widely accepted, this

is due to high chlorine content of these substances. The relationship between ozone depletion potential and global warming potential is the major concern in the field of GRT (green refrigeration technology). As pointed out in the study of Dalkilic and Wongwises [3], ozone depletion potential and global warming potential have become the most important criteria in the development of new refrigerants. This had lead researchers to have started focusing on investigation into more environmental friendly refrigerants than CFCs and HCFCs refrigerants for the protection of the environment such as the use of R134a, ammonia and hydrocarbon (HC) refrigerants as working fluid in refrigeration and air-conditioning systems. Hydrocarbon refrigerants have the advantages of low local availability and environmental acceptability when compared to CFCs and HCFCs. This is due to the fact that, the absence of chlorine in Hydrocarbon (HC) and ammonia results in zero ozone depleting potential (ODP) and has a negligible global warming potential (GWP). The studies according to Kalla and Usmani [4] also provide a review of the efforts to replace the HFCs (hydrofluorocarbons) which are harmful to the environment. The performance of a vapour compression refrigeration system can be improved in a number of ways, other than by testing the system in a controlled environment experimentally; one of those ways is the use of computer models to simulate the thermal and fluid-dynamic behaviour of refrigeration systems to achieve rapid and accurate result that will aid optimum design of the system.

Application of simulation performance prediction and optimum design of refrigeration systems has been pointed out in the study of Guo-liang Ding [5]. Simulation techniques have also been used by researchers for design of vapour compression refrigeration system under steady-state conditions (Sanaye and Malekmohammadi [6]). According to Winkler [7], a steady-state simulation provides details regarding the system performance at a set of design points and describes how the system will perform at off-design conditions. Raghunatha Reddy et al. [8] carried out investigation on application of soft computing techniques for analysis of vapour compression refrigeration system. The result from their study revealed that intelligent systems such as autonomous self-tuning systems and automated designed systems are applicable with the aids of computer simulation techniques. Many investigations have been carried out both experimentally and numerically in search for more eco- friendly refrigerants that could be retrofitted to existing vapour compression refrigeration system so as to minimize adverse impact of thermal systems usage on the environment. Tallita et al. [9] developed a mathematical model using adaptive time step Runge-Kutta Fehlberg fourth-fifth order method to compare the performance of vapour compression refrigeration system substituting R12 for R134a, ammonia and R600a. The simulation results of their study revealed R134a as the best fluid that could be used to replace R12 in vapour compression refrigeration system. Mohan [2] reported comparative analysis of the refrigerant impact on the operation and performances of a one stage vapour compression refrigeration system using R12 and R134a as refrigerants. According to the author, R134a is observed to be more sensitive to variation in degree of sub-cooling from COP evaluation point of view. Hadya et al. [10] studied and compared theoretically the possibility of using R32 and R290 refrigerants with R22 in a lower capacity air-conditioning system. Close match of COP values for R290 and R32 are observed with that of R22 counterpart based on the experimental operating conditions considered in their study.

To prevent the environmental damage and to reduce the harmful effects of the CFCs and HCFCs application in refrigeration industry further investigation are carried out in the present study using computer model to simulate fluid-dynamic behaviour of alternate refrigerants in refrigeration systems. Simulation and comparative analysis of performance of R134a, R290, R600a and ammonia as refrigerants in a single stage vapour compression system was carried out. The performance evaluation of the system in terms of COP, refrigerating effect and overall efficiency of the system values obtained for R134a, HCs and ammonia refrigerants were comparatively analysed with that of CFCs and HCFCs refrigerants.

2. VAPOUR COMPRESSION REFRIGERATION SYSTEM DESCRIPTION

Figure 1 shows schematic diagram of a single stage vapour compression refrigeration system and the corresponding P-h diagram of the system is presented in Fig. 2. This system consists of a compressor, a condenser, expansion valve (capillary tube) and an evaporator. The arabic numbers in Fig. 1 from 1 to 4 show the different state of the vapour compression refrigeration system and the number sequence indicates the flow direction of refrigerant in the system. The refrigerant enters the compressor at state 1, as saturated vapor and also with respect to the evaporation temperature. It follows the irreversible compression process 1-2. At state 2 the refrigerant is with extremely high pressure and superheated. The compressed refrigerant vapor runs from state 2 to state 3. The refrigerant vapor at state 2 influxes into the condenser to be condensed and takes place the heat exchange with the surroundings hence arriving in state 3 as saturated liquid which is further throttled at constant enthalpy during the process 3-4. The cycle is closed by a vaporization process 4-1 in which the refrigerant (two-phase mixture) is evaporated at constant pressure heat interaction in internally reversible condition to a saturated vapor at state 1. The process of the evaporation of refrigerant in the evaporator thus receives heat from the cooling/refrigerated space.

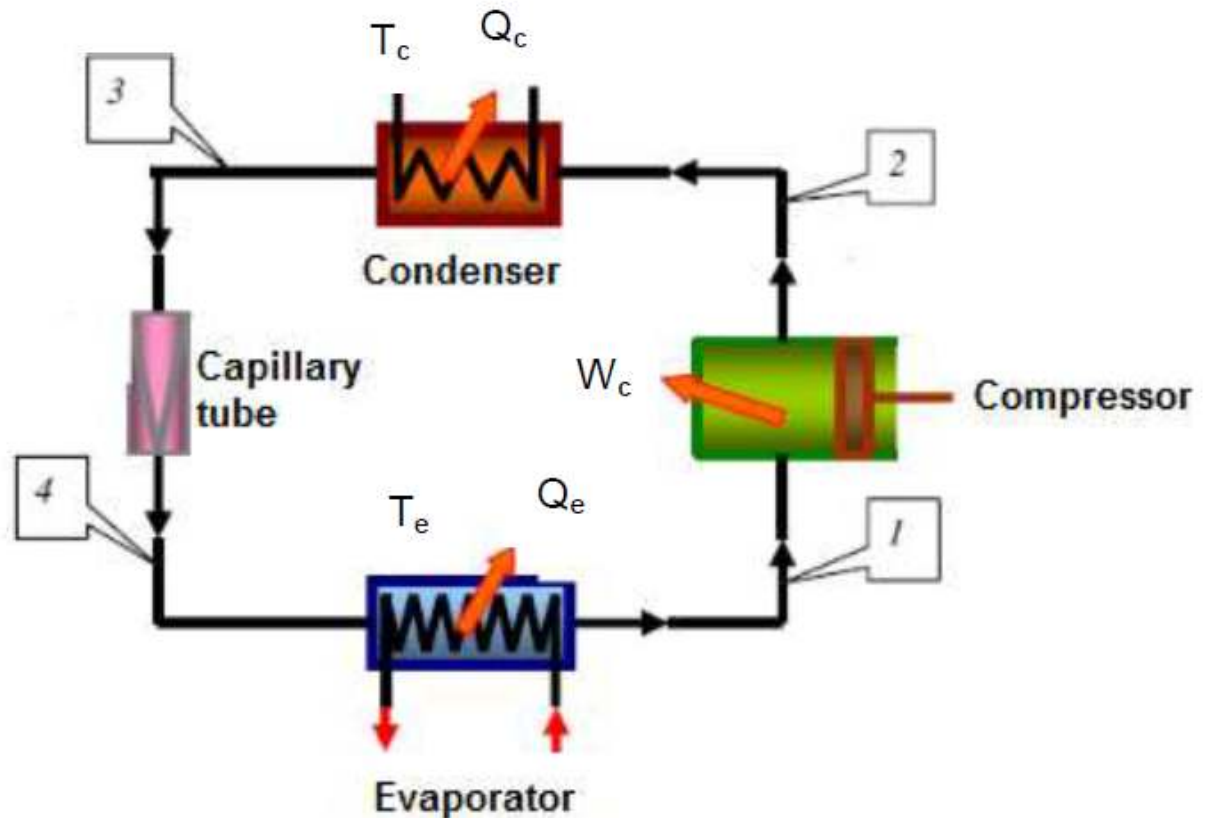


Fig. 1: Single stage Vapour Compression Refrigeration System.

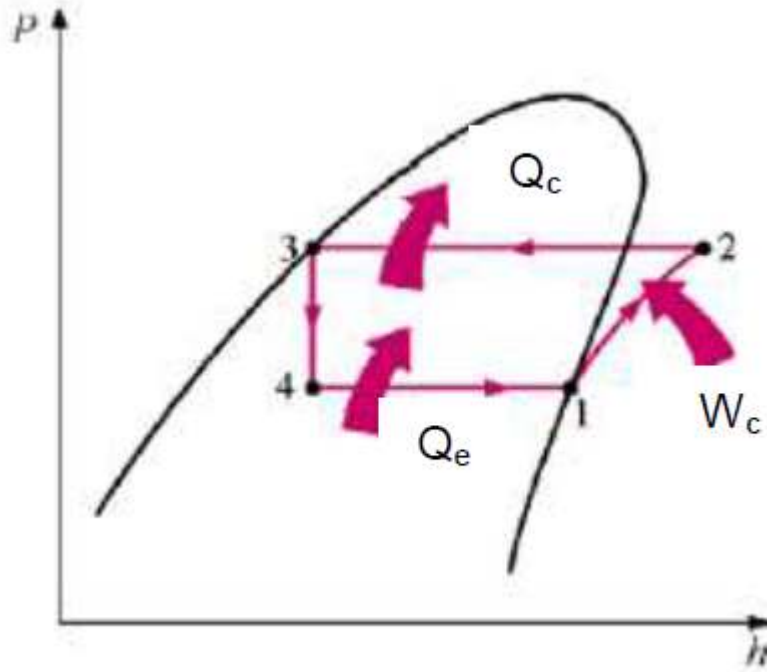


Fig. 2: P-h Diagram Vapour Compression Refrigeration System.

3. MATHEMATICAL MODEL

In this section the system components level mathematical models were developed based on the mass and energy conservation principles described as follows:

3.1 Refrigeration system cooling load model

In the present study, the system cooling load (Q_{CL}) which comprises of both the product load (Q_{PT}) and the infiltration load (Q_I) is modelled as follows:

$$Q_{CL} = Q_{PT} + Q_I \quad (1)$$

where Q_{PT} is the total product cooling load estimated such that for n product stored in the system, the total product cooling load is calculated as:

$$Q_{PT} = \sum_1^n Q_P \quad (2)$$

The term Q_P in Eq. (2) is the total load required to cool a product from storage temperature t_1 to final temperature t_3 and is determined using Eq. (3):

$$Q_p = Q_{AF} + Q_F + Q_{BF} \quad (3)$$

123

124 where the terms Q_{AF} , Q_F and Q_{BF} are defined as sensible heat load above freezing,
 125 latent heat of freezing, and sensible heat load below freezing respectively for the selected
 126 products and they are given as follow:
 127

$$Q_{AF} = mc_a(T_1 - T_2) \quad (4)$$

129

$$Q_F = mh_{fg} \quad (5)$$

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$$Q_{BF} = mc_b(T_2 - T_3) \quad (6)$$

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134 The detail of the terms ($n, m, T_2, c_a, c_b, h_{fg}$) for the selected products considered in the
 135 present study are shown in Table 1.
 136

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Table 1. Properties of the selected products

Products	Mass ,m [Kg]	Highest freezing temperature, T_2 [°C]	Specific heat above freezing, c_a [KJ/KgK]	Specific heat below freezing, c_b [KJ/KgK]	Latent heat of fusion, h_{fg} [KJ/KgK]
Apple	5	-1.5	3.64	1.88	506.26
Fresh meat	5	-1.67	3.18	2.13	418.40
Water	5	-0.3	3.94	2.01	310.24

138 Source: Heat load in refrigeration systems [11]

139 It is interesting to highlights that the average ambient temperature of the
 140 environment T_1 and the final temperature T_3 are assumed to be 30 °C and -10 °C
 141 respectively.

142 The infiltration load (Q_i) given in Eq. (1) is calculated from:

143

$$Q_i = 1.08V(T_1 - T_3) \quad (7)$$

145

146 where 1.08 is a multiplying factor and V is the average velocity of the door and is computed
 147 as:
 148

$$V = v_r \sqrt{\frac{H}{H_r}} \sqrt{\frac{T_a}{T_r}} \quad (8)$$

150

151 where,

152 V_r is the average velocity of the reference door

153 H is the height of the door

154 H_r is the height of the reference door

155 T_a is the temperature difference between the refrigerated space and the environment

156 T_r is temperature difference of the reference door.

157

158 In the present study, the concept of energy balance in thermodynamics cycles,
159 applying first Law of thermodynamics for control volumes to obtain performance result that
160 can meet the operating conditions imposed on each component shown in Fig.1 was adopted
161 and is mathematically expressed according to Eq. (9)

162

163
$$\dot{Q}_{vc} + \sum \dot{m}_o \left(h_o + \frac{v_o^2}{2} + gz_o \right) = \frac{dE_{vc}}{dt} + \sum \dot{m}_i \left(h_i + \frac{v_i^2}{2} + gz_i \right) + \dot{W}_{vc} \quad (9)$$

164

165 Where the subscripts i and o in Eq. (9) stands for inlet and outlet states, respectively.

166 It is well known that in vapour compression refrigeration system, changes in kinetic, $v^2/2$
167 and potential energies, gz are negligible. Thus Eq. (9) becomes:

168

169
$$\dot{Q}_{vc} - \dot{W}_{vc} = \sum \dot{m}_o h_o + \sum \dot{m}_i h_i \quad (10)$$

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171 Referring to the P-h diagram shown in Fig. 2 and application of Eq. (10) to each
172 component of the system, the mathematical equations used to obtain the energy balance in
173 each component are presented in the next subsections:

174

175 3.2 Heat exchangers models: evaporator, condenser

176 The heat absorbed by the evaporator (i.e., cooling capacity) is calculated from:

177
$$Q_e = \dot{m}(h_1 - h_4) \quad (11)$$

178 The heat released by the condenser is computed according to Eq. (12) as follows:

179
$$Q_c = \dot{m}(h_2 - h_3) \quad (12)$$

180

181 3.3 Compressor

182 An overall energy balance applied to calculate the rate of work input to the compressor is
183 expressed mathematically as:

184
$$W_c = \dot{m}(h_2 - h_1) \quad (13)$$

185

186 3.4 Expansion valve

187 For the expansion process, the overall energy balance in the capillary tube is calculated
188 from:

$$\dot{m}h_3 = \dot{m}h_4 \quad (14)$$

The refrigerant mass flow rate ($\dot{m}$) of the systems is calculated from:

$$\dot{m} = \frac{Q_{CL}}{h_1 - h_4} \quad (15)$$

The Mathematical model described above was implemented in MATLAB [1] software to predict the response of principles components i.e compressor, condenser, expansion valve (capillary tube) and evaporator. The simulation model of the single stage vapour compression refrigeration system shown in Fig.1 was devised assuming the following conditions (i) the refrigeration system operates at steady state regime, (ii) irreversibilities within the evaporator, condenser and compressor are ignored, (iii) no frictional pressure drops, (iv) refrigerant flows at constant pressure through the two heat exchangers (evaporator and condenser) , heat loss to the surrounding are ignored and compression process is isentropic. The solution algorithm is illustrated in the information flow diagram depicted in Fig. (3).

The software input data are refrigerant type, product properties, evaporation temperature and condensation temperature. The thermodynamics parameters (P, h, v, s) for the refrigerants considered were set from COOLPACK saturation table [12]. The first simulation of the system was performed using the thermodynamic properties of R12. Subsequently, new simulations were carried out for other refrigerants such as R22, R134a, R600a, R290, and ammonia considering their respective thermodynamic properties. It is interesting to point out that the effect of quantity of products refrigerated on the performance of vapour compression refrigeration system can be evaluated with the developed model taking into consideration Eqns. 1-8 thus, the performance prediction and optimum design of refrigeration systems can be achieved with this model.

Based on the simulation procedure carried out for each refrigerant, the performance of the system is evaluated as follows:

The coefficient of performance (COP) of the system is calculated from:

$$COP_{Ref} = \frac{Q_e}{W_C} \quad (16)$$

The overall efficiency of the system (η) is estimated from:

$$\eta = \frac{COP_{Ref}}{COP_{Carnot}} \quad (17)$$

The term COP_{Carnot} in Eq. (17) is calculated as follow:

$$COP_{Carnot} = \frac{T_e}{T_e - T_c} \quad (18)$$

where, T_e and T_c are the evaporation and condensation temperature respectively.

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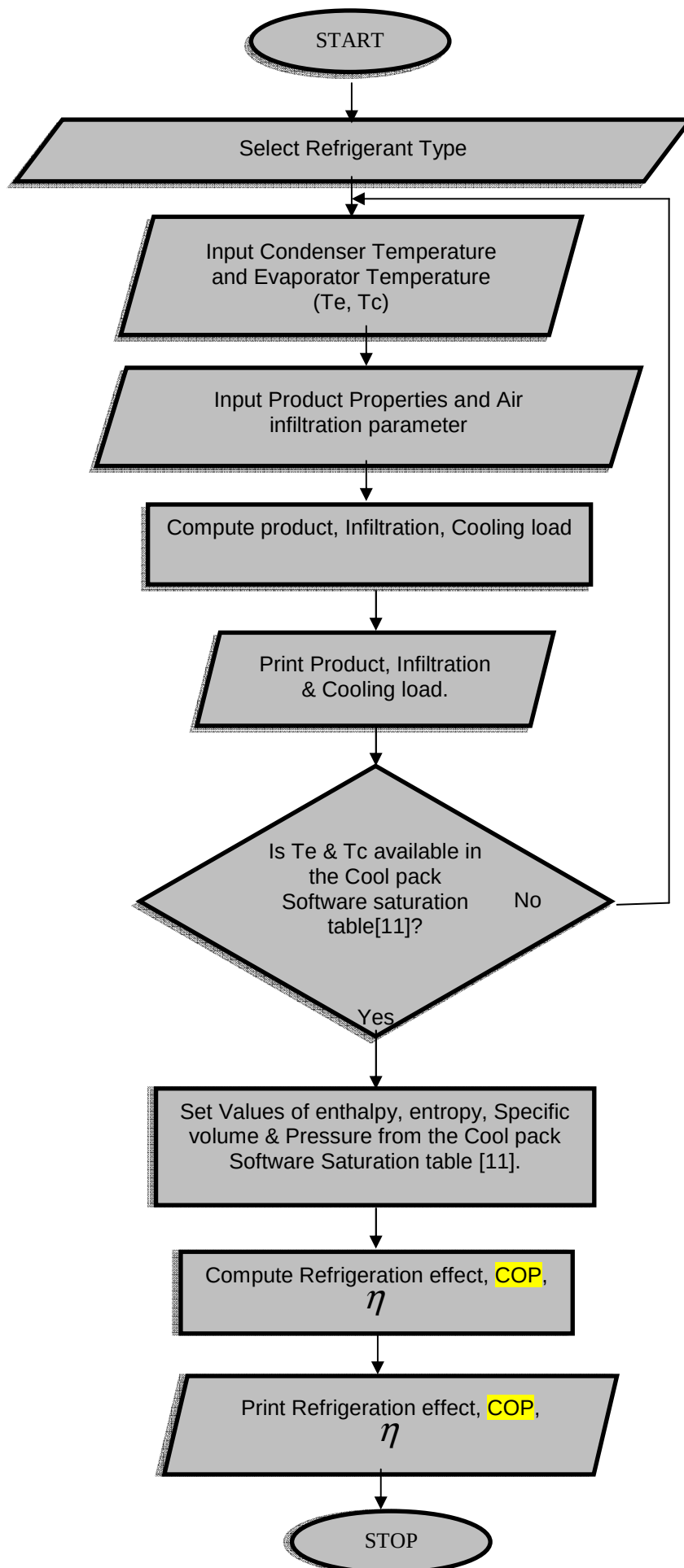


Fig. 3: Flow chart of the simulation program.

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265 4. RESULTS AND DISCUSSION

266 In this section, analysis of performance of HCs, R134a, and ammonia as refrigerants
267 in a single stage vapour compression refrigeration system based on the use of numerical
268 model developed for the system are comparatively reported with those of R12 and R22
269 refrigerants. Simulation data was generated over a wide range of evaporator and condenser
270 temperatures -25 to 15°C and 30 to 60°C respectively for various working fluids such as
271 R12, R22, R134a, R290, R600a and ammonia to observe the performance of the system in
272 terms of refrigerating effect, coefficient of performance (COP) and overall efficient of the
273 system.

274

275 4.1 Comparative analysis

276 An overview of the performance evaluation of the system based on the effects of
277 evaporation and condensation temperatures using the selected refrigerants as working fluid
278 in vapour compression refrigeration system are shown in Figures 4-7. Fig. 4 presents the
279 variation of coefficient of performance with the evaporation temperature at 40°C condenser
280 temperature. Evaluating the Fig. 4, it is clear that the COP which is an important parameter
281 in the analysis of cooling systems like vapour compression refrigeration system increases as
282 expected with increasing the evaporation temperatures of the system. This behaviour is
283 similar to those observed in the study of Mogaji and Yinusa [13]. Close match of COP values
284 for R600a is observed with that of R12 and R22 counterpart under the similar operating
285 conditions considered in the present study. It can also be noticed that for the same capacity
286 compressor, COP of R134a and R290 is less than that of R12 and R22. However, among
287 the refrigerants used in the simulated system under the similar operating conditions, the
288 system with the use of ammonia yields highest COPs and are of up to 6.3 and 7.5%
289 increment compared with those of R12 and R22 respectively.

290 The effects of condensation temperatures on the COP of the system at -10°C
291 evaporator temperature using the selected refrigerants considered in the present study as
292 shown in Fig. 5 revealed that the COP for the system decrease as the condensation
293 temperature increases. This trend is similar to those observed in the studies of Mogaji and
294 Yinusa [13] and Elgendy [14]. From Fig. 5, it can be observed that the COP value was nearly
295 invariable among the systems simulated with the use of R12, R22, R290 and R600a.
296 However, the simulated system with ammonia gave highest COP value as similarly observed
297 in the study of Kilicarslan and Muller [15] and the COP value is of about 6 and 4 at
298 condensation temperatures of 30 and 50°C respectively. These results also revealed that
299 the system with the use of ammonia as a working fluid has the better performance compares
300 with that of R12 and R22 under similar operating conditions. Simulation of the system with
301 the use of R290 and R600a also showed better performance compared with other
302 refrigerants considered in this study from the effect of condensation temperature on the
303 system refrigerating effect point of view as shown in Figs. 6 and 7. It can also be noticed
304 from these figures that the performance of the system with the use of R134a is relatively
305 close to those of the systems with R12 and R22. It is interesting to highlight that the
306 performance evaluation simulation results for ammonia in terms of refrigerating effect though
307 not shown in Figure 6 and 7 is on a high side compared to the other refrigerants considered
308 in this study.

309

310 The overall efficiency results of the simulated system with the use of the selected
311 refrigerants obtained using Eq. (18) by taking into consideration the effects of evaporation
312 and condensation temperatures on the system are presented in Tables 2 and 3. The results
313 contained in Tables 2 and 3 also revealed that the overall efficiency of the system with R12,
R22, R134, R290 and R600a are nearly matching the same values. However, the use of

ammonia under the same operating condition gave the highest overall efficiency of the system. For the ammonia, the simulation resulted in overall efficiency for about 84 and 92% at high evaporation and condensation temperatures of 15 and 50 °C respectively. These results justify the reason for better performance of the system with the use of ammonia among others refrigerants as displayed in Figs. 4 and 5.

The results from this study revealed that R12 and R22 may be replaced by ammonia, R134a, R600a and R290 without any significant loss in the overall performance vapour compression refrigeration system considering the fact that the impact of ammonia on ozone layer (ODP) and global warming potential (GWP) is null but based on its corrosive and toxic nature, it can be used by using safety measure for industrial applications, R134a is easily available in the market and its impact on the ozone layer is negligible compared with that of R12 and R22, also with the use of modern hermetically sealed system in refrigeration system, the problems of flammability posed by the use of hydrocarbons refrigerant are now reduced to some extent. Moreover, among the refrigerants comparatively analysed, the use of ammonia was the best working fluid for the vapour compression refrigeration system simulated in this study. This is because, the simulated system with ammonia presented the best COP and highest overall efficiency of the system compared with other refrigerants considered in this work.

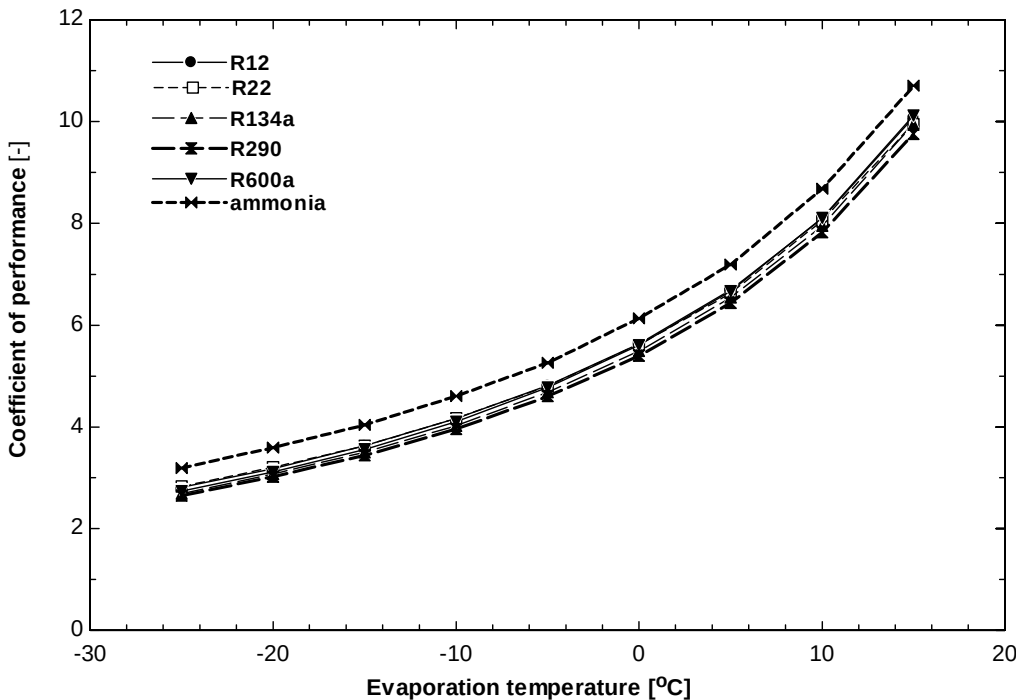


Fig. 4. Variation of coefficient of performance with evaporation temperature

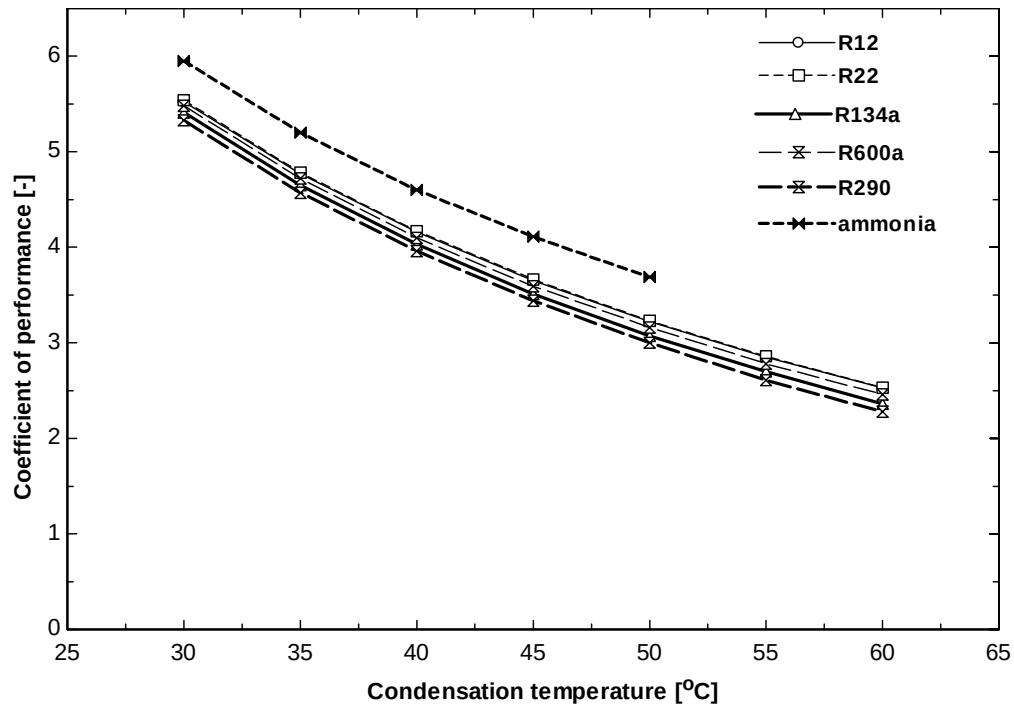


Fig. 5. Variation of coefficient of performance with condensation temperature

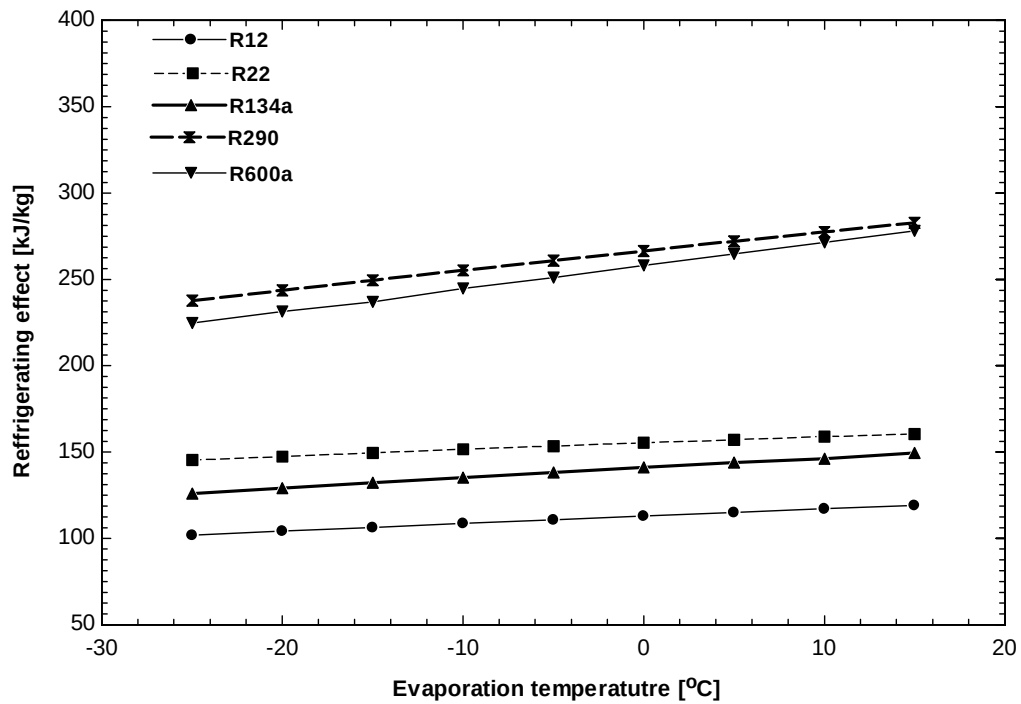


Fig. 6. Variation of refrigerating effect with evaporation temperature

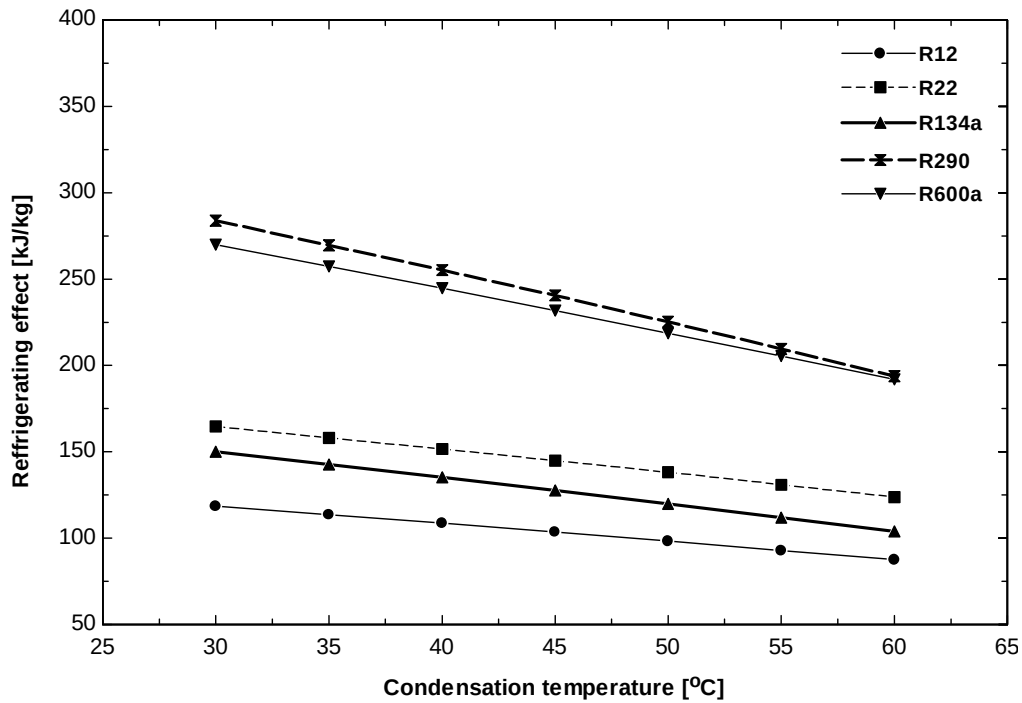


Fig. 7. Variation of refrigerating effect with condensation temperature

Table 2. Effects of evaporation temperatures on the overall efficiency the simulated system.

Tevap [°C]	R12 η [%]	R22 η [%]	R134a η [%]	R290 η [%]	R600a η [%]	ammonia η [%]
-25	73.7	74.2	70.4	69.5	71.9	83.6
-20	75.5	76.1	72.7	71.6	74.0	85.1
-15	77.3	77.5	74.6	73.3	76.0	86.0
-10	79.1	79.3	76.6	75.4	78.0	87.5
-5	80.7	80.5	78.5	77.0	80.0	88.4
0	82.5	82.3	80.5	79.0	82.0	89.8
5	84.2	83.5	82.3	81.1	84.0	90.5
10	85.8	85.4	84.3	82.9	86.0	91.9
15	87.5	86.4	86.3	84.8	88.0	92.6

Table 3. Effects of condensation temperatures on the overall efficiency of the simulated system.

Tcond [°C]	R12 η [%]	R22 η [%]	R134a η [%]	R290 η [%]	R600a η [%]	ammonia η [%]
30	84.0	84.3	82.2	81.4	83.4	90.5
35	81.6	81.9	79.5	78.2	80.8	89.0
40	79.1	79.3	76.6	75.4	78.0	87.5
45	76.4	76.6	73.5	72.0	75.1	85.7
50	73.5	73.7	70.1	68.4	72.1	84.2

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353 5. CONCLUSIONS

354 Analyses of performance of Hydrocarbons (HCs), R134a, and ammonia as refrigerants
355 in a single stage vapour compression refrigeration system based on the use of simulation
356 model developed for the system are comparatively reported with those of R12 and R22
357 refrigerants in this paper. Performance evaluation of the system was characterized in terms
358 of coefficient of performance (COP), refrigerating effect and overall efficiency of the system
359 using the various selected working fluid considered in this work. The effects of condensation
360 and evaporation temperatures were studied on the system operation and performances.
361 From the present study, the following main conclusions can be drawn:

362

- 363 i. The COP and Refrigerating effect of the vapour compression refrigeration system
364 increase with increasing the evaporation temperatures and decreasing the
365 condensation temperatures of the system.
- 366 ii. Close match of COP and the system overall efficiency values for, R134, R290 and
367 R600a and R290 are observed with that of R12 and R22 counterpart under the
368 similar operating conditions.
- 369 iii. Among the refrigerants simulated in the system under the similar operating
370 conditions, the system with the use of ammonia yields highest COPs and are of up
371 to 6.3 and 7.5% increment compared with those of R12 and R22 respectively.
- 372 iv. Ammonia was observed to be the best working fluid for the vapour compression
373 refrigeration system presented the best COP and highest overall efficiency of the
374 system compares with other refrigerants considered in this study.
- 375 v. According to the preceding comparative analysis on performance evaluation of
376 vapour compression refrigeration system. It can be concluded that R12 and R22
377 may be replaced by ammonia, R134a, R600a and R290 without any significant loss
378 in the overall performance of the system provided the safety issues are addressed in
379 terms of manufacturing, handling, storage and servicing so as to prevent
380 flammability and corrosive/toxic nature of hydrocarbons and ammonia respectively.

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