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## Original Research Article

# MORPHOLOGY, FUNCTIONAL AND PASTING PROPERTIES OF GINGER STARCHES PREPARED BY FOUR DIFFERENT DRYING METHODS

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### ABSTRACT

**Aim:** This study aims at providing information on the morphology, physico-chemical and pasting properties of the ginger starches prepared by air drying, oven drying, freeze drying and microwave drying methods with a view to improving their utilization.

**Place and Duration of Study:** Department of Pharmaceutics and Pharmaceutical Technology, Olabisi Onabanjo University, Nigeria and Department of Pharmaceutics, University of Ibadan, Nigeria between March 2013 and February 2014

**Methodology:** Starch was isolated from the rhizomes of *Zingiber officinale* and dried by oven, freeze drying, air drying and microwave drying methods. They were studied for their composition, morphological, functional and pasting properties

**Result:** The residual protein content of the four starches was slightly different averaging 5.06 %. The residual protein of air dried starch was the lowest and that freeze dried starch was the highest. There was no significant variation in the saponin and tannin contents of the dried samples. The amylose content of the ginger starches ranged between 20.7% and 22.1%. The rank order of the amylose content was microwave dried > freeze dried > air dried > oven dried. Drying methods had a significant effect ( $P < 0.0001$ ) on the solubility index and swelling capacity of the dried samples investigated. The rank order of the swelling capacity of the ginger starches was freeze dried < air dried < oven dried < microwave dried. A higher solubility index was observed among the oven-dried sample and the microwave sample compared to freeze dried and air dried samples. X-ray fluorimetric (XRF) analysis of the different starch powders showed the presence of iron, copper and zinc in trace amounts; and the absence of heavy metals like lead and mercury. All the four dried samples have fairly good flow. Significant differences were observed in individual pasting parameters of the ginger starches especially in peak viscosity, trough viscosity, final viscosity and setback viscosity. Peak viscosity was found to be lowest for freeze dried starch (300.42RVU) and highest for oven dried starch (324.25 RVU). The rank order of the final viscosity was oven dried > microwave dried > air dried > freeze dried.

**Conclusion:** This study revealed that the physicochemical as well as the pasting properties of the ginger starches prepared by the four drying methods make them to be considered as excellent resource with possible applications in many food and pharmaceutical processing. The results also showed that difference in drying methods during processing has an effect

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*Keywords: Ginger starch; morphology; functional; pasting properties; drying methods*

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## 1. INTRODUCTION

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Starch, the main reserve carbohydrate of several crops, is highly abundant in nature and can be easily extracted with high purity and low cost. Industries have been using starch for a long time as an ingredient in foods, especially for its functional properties. Depending on the source, starches have different applications in the food and pharmaceutical industries where they are used as fillers, glidants, thickeners, binders, disintegrants as well gelling, bulking and water retention agents [1]. Starch which is a major dietary source of carbohydrate, is the most abundant storage of polysaccharide in plants, and occurs as granules in the chloroplast of green leaf and the amyloplast of seeds, tubers and rhizomes. Starch consist of two molecules- the linear and helical amylose, and the highly branched amylopectin. They usually consist of 10-38 % amylose although the proportion often varies depending on the botanical source of the starch [2]. Tropical roots and tubers, grains, cereals and fruits are crops that have served as staple foods for millions of people throughout the world for many centuries and they generally have a high starch **content** which has made them potential sources of industrial starch.

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Ginger is a perennial creeping plant, with thick tuberous rhizome, producing an erect stem 30 - 100 cm (1-3 ft) tall. The lance-shaped leaves are bright green, 15 - 20 cm (6-8 in) long, with a prominent longitudinal rib, enclosing conical clusters of small yellow-green flowers marked with purple speckles. It is propagated from rhizome cuttings, planted on rich, well drained loam. The underground rhizomes are the medicinal and culinary useful part of the plant. Ginger is widely used as a flavoring agent in beverages and many food preparations. Ginger helps relieve indigestion, gas pains, diarrhoea and stomach cramping. The primary known constituents of ginger root include starch, gingerols, zingibain, bisabolene, oleoresins, essential oil (zingiberene, zingiberole, camphene, cineol, and borneol), mucilage, and protein.

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Starch isolated from ginger rhizomes has been investigated for its use as pharmaceutical excipients [3-6] **but it appears little or no work has been done to improve their** functional properties and enhance their industrial uses in Pharmacy as thickener, binder, filling material and encapsulating agent. The increased requirements for starches in different applications, such as functional and healthier food, or applications in pharmaceuticals have thus forced a steady development of new starch types and the crucial need to understand the properties of the existing ones. This study was conducted to provide information on the morphology, physico-chemical and pasting properties of the ginger starches prepared by air drying, oven drying, freeze drying and microwave drying methods with a view to improving their utilization.

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## 2. MATERIAL AND METHODOLOGY

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### 2.1 MATERIALS

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Ginger starch was extracted from ginger rhizome at the Department of Pharmaceutics and pharmaceutical technology laboratory of Olabisi Onabanjo University, Nigeria. Water was double distilled and all other chemicals were of analytical grade

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### 2.2 METHODOLOGY

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#### 2.2.1 Extraction of Ginger starch from ginger rhizome

64 Rhizomes of ginger (*Zingiber officinale* Roscoe, Family Zingiberaceae) was washed with  
65 water, peeled and grated to produce ginger slurry. The grating was intermittently done to  
66 prevent the starch from heating up due to heat from grater. Enough quantity of distilled water  
67 was added to soak the material for 5 hours. The slurry was then thoroughly washed with  
68 water onto the muslin cloth into a collecting vessel to release the starch granules embedded  
69 in the parenchyma cells. The content of the collecting vessel was the allowed to settle for 2  
70 hrs and the yellowish supernatant was decanted. Series of re-dispersion and decanting were  
71 done to remove impurities. The settled starch was scrapped off and divided into four equal  
72 parts, each of which was subjected into different drying methods which are; freeze drying,  
73 oven drying, microwave drying and open air drying.

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## 75 **2.2.2 Determination of Functional properties**

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### 77 *2.2.2.1 Determination of bulk and tapped density*

78 This was carried out for each sample using granules of mesh sizes of 500 – 1000 microns. A  
79 50g sample was poured into a 50ml measuring cylinder and the bulk volumes were noted.  
80 The bulk density was calculated from the ratio of the mass to the bulk volume. The  
81 procedure was repeated and result obtained in triplicate. Tapped density determination was  
82 carried out by tapping the cylinder 100 times and the tapped volume obtained was used to  
83 calculate the tapped density.

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### 85 *2.2.2.2 Determination of particle density*

86 The particle densities of the dried ginger starch powders were determined by the  
87 pycnometer method using liquid immersion technique with benzene as the displacement  
88 liquid. A 50 mL pycnometer bottle was weighed when empty (w). This was filled with  
89 benzene to the brim till it overflows. The excess was wiped off the bottle and its contents  
90 were weighed (W1). The difference between the two weights recorded was calculated (W2).  
91 A 2 g quantity of the ginger starch powder was weighed (W3) and quantitatively transferred  
92 into the pycnometer bottle. The excess solvent was wiped off and the bottle weighed again  
93 (W4). The particle density,  $\rho$  (g/cm<sup>3</sup>), was calculated from the following:

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$$\rho = \frac{W2 \cdot W3}{50 (W1+W3-W4)} \dots\dots\dots (1)$$

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### 97 *2.2.2.3 Determination of flow properties*

98 A 30 g sample of the powders was poured into a plugged glass funnel with the tip, 10 cm  
99 above the flat surface of the bench. The granules were allowed to flow freely through the  
100 orifice of the funnel to form a heap whose height and diameter were determined. The angle  
101 of repose was calculated using the equation below:

$$\text{Tan } \theta = h/r \dots\dots\dots (2)$$

103 Where h = height and r = radius of circular heap

104 The hausners ratio was calculated as the ratio of the tap density to the bulk density of the  
105 samples while the compressibility Index (C %) was calculated using bulk and tap densities  
106 data when fitted into the equation below:

$$\text{Compressibility index} = (\text{tapped density} - \text{bulk density}) / \text{tapped density} \times 100 \% \dots\dots\dots (3)$$

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### 109 *2.2.2.4 Determination of Swelling capacity and solubility Index*

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110 The method described by Leach et al., 1959 [7] was used with slight modifications. Starch  
111 blends (1 g) were weighed and transferred into a clean dry test tube and weighed (W1). The  
112 mix was then dispersed in 50 mL of distilled water using a magnetic stirrer. The resulting  
113 slurry was heated at 80°C for 30 minutes in a thermo stated water bath. The mixture was  
114 cooled to room temperature and centrifuged at 15 minutes. 5 mL of aliquot of the  
115 supernatant was dried to a constant weight at 120°C. The residue obtained after drying  
116 represented the amount of starch solubilised in water. Solubility was calculated as per 100g  
117 of starch on dry basis. The residue obtained after centrifugation with the water it retained  
118 was transferred to the clean dried test earlier and re-weighed (W2).

119 Swelling of starch =  $(W2 - W1) / \text{weight of powder} \times 100$  ..... (4)  
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#### 121 2.2.2.5 Determination of moisture content

122 A 5g powder was weighed in a tarred crucible and dried in hot air oven (Mettler Model  
123 200) at 100±5°C till a constant weight is obtained. The moisture contents were calculated by  
124 the formula given below. The mean of three determinations was taken as the final moisture  
125 content

126  $MC = W_0 - W_1 / W_0$ ..... (5)

127 Where MC is the moisture content and  $w_0$ ,  $w_1$  are initial and final weights of the starch  
128 respectively.

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#### 130 2.2.2.6 Determination of protein, tannin and saponin content

131 The protein and tannin content of the dried ginger starch samples were determined by  
132 established methods [8, 9]

#### 133 2.2.2.7. Analysis of elemental constituents

134 Elemental analysis was carried out by the methods of Mineral Analysis using Atomic  
135 Absorption Spectrophotometer. Weight was recorded to the nearest 0.001g

#### 136 2.2.2.8. Fourier Transform Infrared Spectroscopy (FT-IR)

137 The FTIR spectrum of the starches was recorded with a Perkin Elmer RXI  
138 spectrophotometer (Connecticut, USA). The dry powder was mixed with KBr and pressed  
139 into pellets. The spectrum was obtained by scanning between 4000 and 500/cm.

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### 141 **2.2.3 Determination of Pasting properties**

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143 Pasting characteristics were determined with a Rapid Visco Analyser (RVA Super 3,  
144 Newport Scientific Pty. Ltd, Australia). A 3 g sample was dissolved in 25 ml of water in a  
145 sample canister. The sample was thoroughly mixed and fitted into the RVA as recommended  
146 (Newport Scientific, 1998). The slurry was heated from 50 to 95°C with a holding time of 2  
147 min followed by cooling to 50°C with another 2 min holding time. The 12 min profile was  
148 used and the rate of heating and cooling was at a constant rate of 11.25°C/min.  
149 Corresponding values for peak viscosity, trough, breakdown, final viscosity, setback, peak

150 time and pasting temperature from the pasting profile were read from a computer connected  
151 to the RVA.

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### 154 **3. RESULTS AND DISCUSSION**

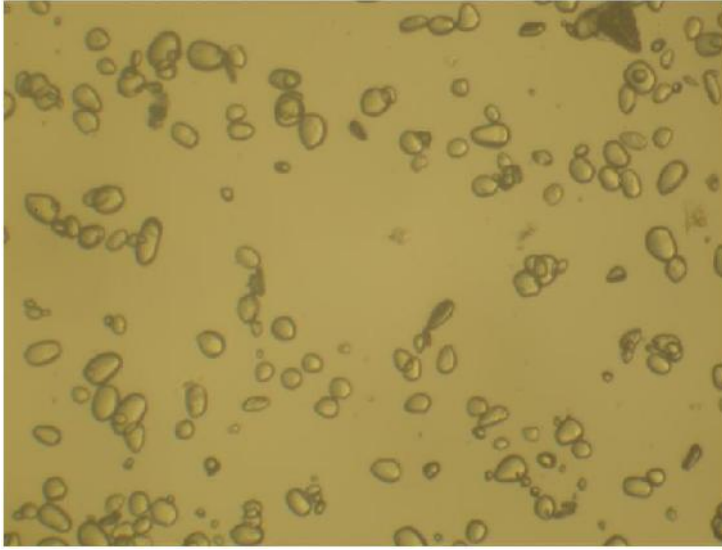
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#### 156 **3.1 Morphology and Composition of ginger starches**

157 The diversity of the shapes of the granules in ginger starches is shown in Figure 1-4. The  
158 shapes range from polyhedral/rounded/oval in air dried sample;  
159 irregular/orbicular/oval/reniform in freeze dried starch; oval/ellipsoidal/rounded in microwave  
160 dried starch and being oval/cordate/polyhedral in oven dried starch. Table 1 presents the  
161 composition and functional properties of the ginger starches obtained from the four drying  
162 method. The residual protein content of the four starches was slightly different averaging  
163 5.06% (Table 1). The residual protein of air dried starch was the lowest while freeze dried  
164 starch had the highest. There was no significant variation in the saponin and tannin contents  
165 of the dried samples. The amylose content of the ginger starches ranged between 20.7%  
166 and 22.1%. The rank order of the amylose content of the starches was microwave dried >  
167 freeze dried > air dried > oven dried. The amylose portion of the starch affects its swelling  
168 and hot-paste viscosities. Shimelis and Rakshit (2005) stated that as the amylose content  
169 increases, the swelling tends to be restricted and the hot paste viscosity stabilizes. Based on  
170 the classification of starches by amylose content [11], all the four dried samples are  
171 intermediate (20-25% amylose).

#### 172 **3.2 Functional properties of ginger starches**

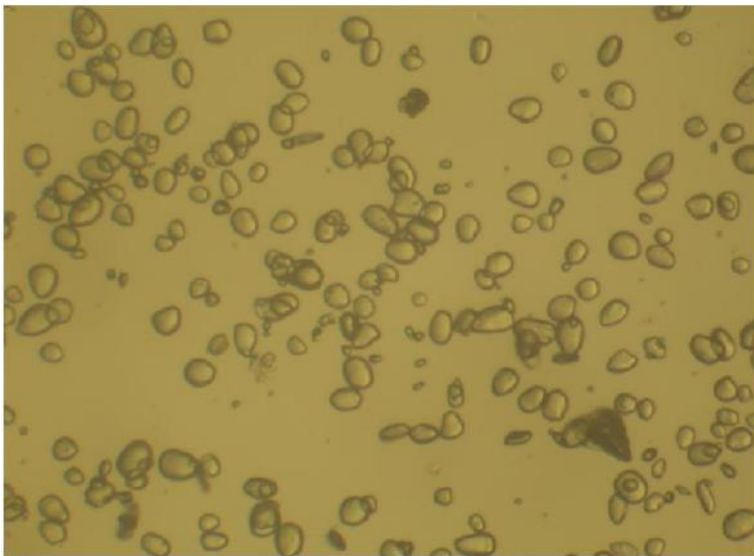
173 Swelling capacity and solubility index provide evidence of the magnitude of interaction  
174 between starch chains within the amorphous and crystalline domains and also evidence of  
175 association bonding within the granules of starches [12]. Drying methods had a significant  
176 effect ( $P < 0.001$ ) on the solubility index and swelling capacity of the dried samples  
177 investigated. The difference in swelling capacity among starches from different drying  
178 methods indicates variation in the strength of associative bonding forces within the granules  
179 [7]. The rank order of the swelling capacity of the ginger starches was freeze dried < air  
180 dried < oven dried < microwave dried. The highest swelling capacity shown by microwave  
181 ginger starch might be indicative of weak bonding forces within its granules and the fact that  
182 it is less compact when compared to the other starch granules. The difference in amylose  
183 content and starch granular properties may also have affected the swelling capacity and  
184 solubility of the ginger starches. A low swelling capacity observed in the freeze dried sample  
185 may be due to stronger bonding force in its starch granules. A higher solubility index was  
186 observed among the oven-dried sample and the microwave sample compared to freeze  
187 dried and air dried samples. High solubility indices in starches has been attributed to the  
188 easy solubility of the linear fraction (amylose) which is linked loosely with the rest of the  
189 macro molecular structure, and released or leached out during the swelling process [18].  
190 Microwave dried sample with the highest amylose content had the highest solubility index  
191 and swelling capacity. Starch granules become increasingly susceptible to shear  
192 disintegration as they swell and starches with lower amylose content swell more than those  
193 with higher amylose content.



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195 **Figure 1: Photomicrography of freeze dried sample x10 magnification**

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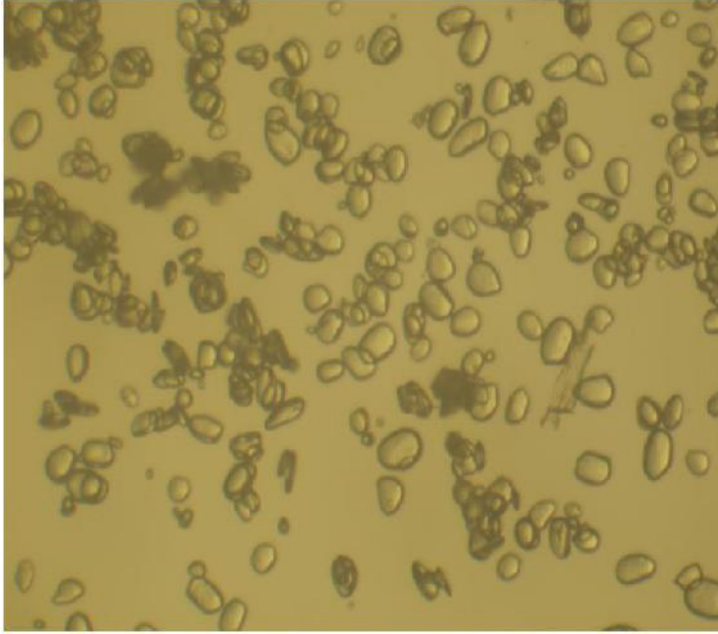
199 **Figure 2: Photomicrography of oven dried sample x 10 magnifications**

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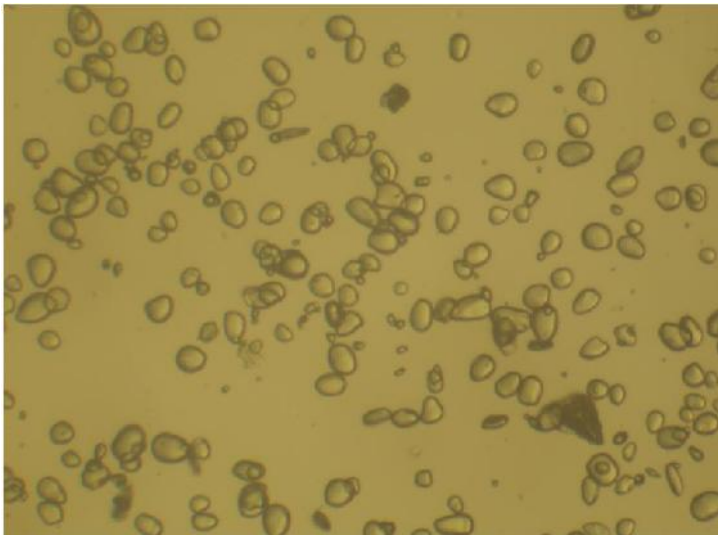
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205 **Figure 3: Photomicrography of microwave dried sample x 10magnifications**

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208 **Figure 4: Photomicrography of air dried sample x10magnification**

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212 **Table 1: Functional properties of dried samples**

| Drying Methods  | Bulk Density (g/cm <sup>3</sup> ) | Tapped Density (g/cm <sup>3</sup> ) | Particle Density (g/cm <sup>3</sup> ) | Angle of Repose (°) | Carrs Index (%) | Swelling Capacity | Moisture Content (%) | Solubility (%) | Protein content | Tannin content (mg/100g) | Saponin content (mg/100g) | Amylose content (%) |
|-----------------|-----------------------------------|-------------------------------------|---------------------------------------|---------------------|-----------------|-------------------|----------------------|----------------|-----------------|--------------------------|---------------------------|---------------------|
| Freeze dried    | 0.418                             | 0.589                               | 1.420                                 | 41.3                | 29.25           | 10.57             | 10.7                 | 9.87           | 4.725           | 6.030                    | 0.125                     | 21.17               |
| Oven dried      | 0.400                             | 0.584                               | 1.420                                 | 43.3                | 31.51           | 10.86             | 10.3                 | 10.17          | 5.270           | 5.850                    | 0.155                     | 20.67               |
| Microwave dried | 0.383                             | 0.563                               | 1.420                                 | 42.7                | 31.97           | 10.93             | 10.8                 | 10.28          | 4.885           | 5.870                    | 0.135                     | 22.06               |
| Air dried       | 0.383                             | 0.563                               | 1.420                                 | 42.3                | 31.97           | 10.65             | 10.6                 | 9.97           | 5.345           | 6.310                    | 0.145                     | 21.14               |

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222 FTIR is a powerful technique to examine the formation of inter- and intra- molecular  
 223 hydrogen bonds in starches. Fourier Transform Infrared Spectroscopy revealed the  
 224 presence of O-H and C-O absorption bands in all the samples. Freeze dried sample had O-  
 225 H and C-O absorption bands distinct at 3350.38 cm<sup>-1</sup> and 1061.52 cm<sup>-1</sup> respectively. Oven  
 226 dried sample had 16 peaks while the air dried sample had 18 peaks. The British  
 227 Pharmacopoeia, 2005 gives limit tests for a number of possible contaminants in  
 228 pharmaceutical raw materials which may be introduced into the finished product during  
 229 processing. Such tests include those for lead, arsenic, calcium, iron, potassium, aluminum,  
 230 halogens and a host of others.

231 **Table 2: Elemental constituent of dried samples**

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| Elements | Freeze dried | Oven dried | Microwave dried | Air dried |
|----------|--------------|------------|-----------------|-----------|
| Fe       | 0.195        | 0.215      | 0.235           | 0.185     |
| P        | 121.055      | 112.260    | 109.46          | 120.145   |
| Ca       | 5.230        | 5.345      | 5.175           | 6.350     |
| Na       | 1.680        | 1.490      | 1.750           | 1.635     |
| Mg       | 3.470        | 2.860      | 3.240           | 3.770     |
| Zn       | 0.435        | 0.435      | 0.415           | 0.415     |
| Mn       | 31.055       | 33.245     | 29.430          | 36.295    |
| Al       | 0.000        | 0.000      | 0.000           | 0.000     |
| K        | 21.470       | 20.45      | 23.59           | 24.230    |
| Pb       | 0.000        | 0.000      | 0.000           | 0.000     |
| Nickle   | 0.000        | 0.000      | 0.000           | 0.000     |
| Selenium | 0.000        | 0.000      | 0.000           | 0.000     |
| Arsenic  | 0.000        | 0.000      | 0.000           | 0.000     |
| Silicon  | 0.000        | 0.000      | 0.000           | 0.000     |
| Cu       | 0.003        | 0.001      | 0.002           | 0.000     |

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234 Although the pharmacopoeial requirements are not categorical on the exact tolerable level of  
 235 any possible contaminant, it should not be presumed that unusual impurities are tolerated.  
 236 X-ray fluorimetric (XRF) analysis of the different starch powders (Table 2) showed the  
 237 presence of iron, copper and zinc in trace amounts. Heavy metals like lead and mercury  
 238 were absent. The presence of heavy metals in formulated products is highly undesirable as  
 239 they form stable covalent or co-ordinate complexes with body protein and can also act as  
 240 catalyst (due to their variable valency state) to induce auto -oxidative reactions. The  
 241 Pharmacopoeia has therefore placed strigent limits on the amount of lead and other heavy  
 242 metals that may be present in pharmaceutical products [14]. Angle of repose, compressibility  
 243 index and Hausner ratio determines the flow character or properties of powder particles. For  
 244 an excellent flow, angle of repose must fall within the range of 25-30 degrees. All the four  
 245 dried samples have fairly good flow which can be improved by the addition of a glidant if they  
 246 are to be used in tablet formulation and production

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252 **Table 3: Pasting properties of Ginger starch powders**

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| Drying Methods  | Peak (RVU) | Trough/Hold (RVU) | Final Viscosity (RVU) | Breakdown (RVU) | Setback (RVU) | Peak Time (min.) | Pasting Temp (°C) |
|-----------------|------------|-------------------|-----------------------|-----------------|---------------|------------------|-------------------|
| Freeze dried    | 300.42     | 151.75            | 222.00                | 148.67          | 70.25         | 5.02             | 82.45             |
| Oven dried      | 324.25     | 200.00            | 290.33                | 124.25          | 90.33         | 4.66             | 81.48             |
| Microwave dried | 306.42     | 163.42            | 236.00                | 143.00          | 72.58         | 4.58             | 82.66             |
| Air dried       | 322.33     | 159.83            | 235.92                | 162.55          | 76.09         | 4.75             | 82.35             |

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255 **3.3 Pasting properties of ginger starches**

256 Table 3 highlights the pasting properties of the starches. Pasting temperature of different  
257 ginger starches ranged from 81.5 to 82.7°C; the highest for microwave starch and the lowest  
258 for oven dried starch. Peak viscosity is the maximum viscosity attained by gelatinized starch  
259 during heating in water. It indicates the water binding capacity of the starch granule. Peak  
260 viscosity was found to be lowest for freeze dried starch and highest for oven dried starch.  
261 The rank order of the peak viscosity was oven dried > air dried > microwave > freeze dried.  
262 Trough viscosity was found to be the lowest for freeze dried starch (200.0 RVU) and the  
263 highest for oven dried starch (151.75RVU). Holding strength measures the ability of starch  
264 to remain undisrupted when paste is subjected to a long duration of high, constant  
265 temperature during the process of steaming. Oven dry method produced starch with the  
266 highest holding strength. Setback is a measure of re-crystallization of gelatinized starch  
267 during cooling. There are significant differences in the values of setback viscosity for the  
268 dried starches. The difference in setback among different starches may be due to the  
269 amount and the molecular weight of amylose leached from the granules of the gelatinized  
270 starch [15]. Oven dried sample had the highest final viscosity. Final viscosity indicates the  
271 ability of the starch to form a viscous paste. The rank order was oven dried > microwave  
272 dried > air dried > freeze dried sample. The increment in the final viscosity of paste made  
273 from oven-dried sample during cooling indicates formation of a firm gel after cooling, rather  
274 than a viscous paste as in the case of other-dried sample. The values of breakdown  
275 viscosity of the starch samples vary significantly. Breakdown viscosity is a measure of the  
276 vulnerability or susceptibility of the cooked starch to disintegration. The higher the  
277 breakdown in viscosity, the lower the ability of the starch sample to withstand heating and  
278 shear stress during cooking [16]. Therefore, oven dried starch (124.25 RVU) and microwave  
279 (143.0 RVU) starches might be able to withstand more heating and shear stress compared  
280 to starches from freeze dried starch (148.67 RVU) and air dried sample (162.55 RVU)  
281 because of their lower breakdown value. Starches with high paste stability or breakdown  
282 have very strong cross-linking within the granules [17]. It therefore means that there is  
283 stronger cross-linking within the granules of air-dried starch powder compared to the other  
284 dried samples.

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286 **4. CONCLUSION**

287 The pasting and physicochemical properties obtained indicate that all the starches have  
288 useful technological properties for many applications. The data showed that the ginger

289 starch prepared by oven drying can be used as a functional ingredient in food systems. This  
290 study revealed that the physicochemical as well as the pasting properties of the ginger  
291 starches prepared by the four drying methods make them to be considered as excellent  
292 resource with possible applications in many food and pharmaceutical processing.

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## 294 **AUTHORS' CONTRIBUTIONS**

295

296 Author BOA and BLG designed the study and wrote the protocol. Author BOA, BLG and FY  
297 managed the literature search and performed the statistical analysis. Author BLG, BOA and  
298 FY managed the analyses of the study while author BLG wrote the first draft of the  
299 manuscript. All the authors read and approved the final manuscript.

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## 301 **REFERENCES**

302 1. Singh N, Singh J, Kaur L, Sodhi NS, Gill BS. Morphological, thermal and rheological  
303 properties of starch different botanical sources. *Food Chem.* 2003; 81: 219-231

304 2. Hoover R. Composition, molecular structure, and physicochemical properties of tuber and  
305 root starches, a review. *Carbohydrate Polymers* 2001; 45: 253-267.

306 3. Shok M, Kunle OO, Abdurahman EM. Ginger starch as tablet binder and disintegrant.  
307 *Pharm. World J.* 1992;9: 23-25

308 4. Ibezim EC, Ofoefule SI, Omeje EO, Onyishi VI, Odoh UE. The role of ginger starch as a  
309 binder in acetaminophen tablets. *Sci. Res. Essay.* 2008; 3: 946-950

310 5. Odeku OA, Odeniyi MA, Ogunlowo GO. Influence of ginger and banana starches on the  
311 mechanical and disintegration properties of chloroquine phosphate tablets. *Asian Pac. J.*  
312 *Trop. Med.* 2009;2:13-18

313 6. Prasanna A. Preparation and Evaluation of Combination Tablet Containing Paracetamol  
314 and Ginger Powder and Its Extract. *Res Rev J Pharm Nanotech.* 2013; 1(1): 7-11

315 7. Leach HW, Mc Cowen LD, Schoch TJ. Structure of the starch granules. In swelling and  
316 solubility patterns of various starches. *Cereal Chemistry.* 1959; 36: 534 – 544.

317 8. HACH. Procedures Manual. Systems for Food, Feed and Beverage Analysis. Hach  
318 Company World Headquarters , Colorado. 1990

319 9. Kardel M, Taube F, Schulz H, Schütze W, Gierus M. Different approaches to evaluate  
320 tannin content and structure of selected plant extracts – review and new aspects. *J Appl Bot*  
321 *Food Qual.* 2013; 86: 154 - 166 (2013), DOI:10.5073/JABFQ.2013.086.021

322 10. Shimelis AE, Rakshit SK. Proximate composition and physico-chemical properties of  
323 improved haricot bean (*Phaseolus vulgaris* L.) varieties grown in Ethiopia. *J. Food Sci.*  
324 *Technol (LWT)*2005; 38: 331-338.

325 11. Juliano BO. Structure and function of the rice grain and its fractions. *Cereal Foods*  
326 *World.* 1992; 37: 772-774.

- 327 12. Jimoh KO, Olurin TO, Aina JO. Effect of drying methods on the rheological  
328 characteristics and colour of yam flours. African Journal of Biotechnology. 2009; 8 (10):  
329 2325-2328.
- 330 13. British Pharmacopoeia, Vol. II and IV. 2005. Her Majesty's Stationery Office, London.  
331 2184-2186
- 332 14. United State Pharmacopoeia and National formulary 2006. Webcom Limited, Toronto,  
333 Asian Edition. 3259-61
- 334 15. Loh J. The effect of shears and strain on pasting behavior of food starches. Journal of  
335 Food Engineering. 1992; 16: 75-89.
- 336 16. Adebowale AA, Sanmi LO, Awonorin SO. Effect of texture modifiers on the  
337 physicochemical and sensory properties of dried fufu. Food Science Technology  
338 International. 2005; 5: 373-382.
- 339 17. Oduro I, Ellis WO, Argeetaoy SK, Ahenkora K, Otoo JA. Pasting characteristics of starch  
340 from new varieties of sweet potato. Tropical Science. 2000; 40 (1):25-28.
- 341 18. Soni RL, Sharma SS, Dun D, Gharia MM, Ahmedabad J. Physico-chemical properties of  
342 Quercus leucotrocophora (oak) Starch/Starke 1993; 45 (4): 127-130

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