Electronic structure with rovibrational and dipole moment calculations of the LaS molecule

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The potential energy and the dipole moment curves have been investigated for the 23 lowest electronic states in the $^{2s+1}\Lambda^{(\pm)}$ representation of the molecule LaS via CASSCF. Multireference CI calculations (single and double excitations with Davidson corrections) were performed by using Gaussian basis sets for the two considered atoms. The harmonic frequency ω_{e} , the internuclear distance r_{e} and the electronic energy with respect to the ground state T_{e} have been calculated for 23 electronic states along with the rovibrational data E_{v} and B_{v} . Nineteen electronic states have been studied here for the first time. The comparison of the investigated values with the theoretical and experimental results available in the literature shows a very good agreement

Keywords: ab initio calculation, spectroscopic constants, potential energy curves, dipole

moment, rovibrational calculation)

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18 1. INTRODUCTION

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20 Diatomic transition metal sulfides molecules are of astrophysical interest. They have also 21 potential applications in chalcogenide based pigments and optical fibers [1-3]. Despite its importance and its various applications [4], the diatomic transition metal sulfides could still be 22 ranked among the least studied diatomic transition metal system [5]. Hinkle et al. [6] called 23 24 for laboratory search for the LaS molecule because of its presence in stellar spectra and S stars [7]. Macrano and Barrow [8] recorded five bands of the $B^{2}\Sigma^{+}X^{2}\Sigma^{+}$ transition of LaS. 25 They produced the spectrum by using a high-temperature carbon tube furnace and observed 26 27 it with a diffraction grating spectrograph. The spectrum showed five four-headed bands 28 degraded to the red in the region 700-850 nm. From their data analysis, they predicted that the origin of the $A^2\Pi_r X^2\Sigma^+$ system of LaS should be at approximately 1100 nm. By using the 29 Los Alamos Fourier transform spectrometer Winkel et al. [9] recorded the spectrum of 30 diatomic lanthanum sulfide. The region 7500-16000 cm⁻¹ is rich in complex bands that were 31 32 identified as the A-X and B-X transitions. A rotational analysis of both the $A^2\Pi - X^2\Sigma^+$, $B^2\Sigma^+$ - $X^{2}\Sigma^{+}$ and a new $A^{2}\Pi - A'^{2}\Delta$ transitions has been performed by Anderson *et al.* [10] where the 33 vibrational and rotational constants have been calculated for the states $X^{2}\Sigma^{+}$, $B^{2}\Sigma^{+}$, $A^{2}\Pi$, and 34 $A'^{2}\Delta$. The (0,0) band of the $B^{2}\Sigma^{+}-X^{2}\Sigma^{+}$ transition of LaS near 726 nm has been studied by He 35 et al. [11] at high resolution using laser vaporization/reaction supersonic free jet expansion 36 and laser induced fluorescence spectroscopy. A total of 1850 rotational-hyperfine transition 37

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lines have been assigned and measured. Accurate rotational, spin and hyperfine parameters for the $X^2\Sigma^+$, $B^2\Sigma^+$ states have been obtained.

The advances in new methods to cool and manipulate neutral molecules offer new possibilities for precision tests of fundamental physics theories. In general relativity the equivalence principle postulates that the outcome of any non-gravitational experiment is independent of position and time. In theories that attempt to unify gravity with other fundamental forces, on the other hand, violation of the equivalence principle may occur and can be consistently described [12]. Precision molecular spectroscopy is a new and promising direction of search for variation of fundamental constants. Molecular spectra are sensitive to

47 both $\mu = m_p/m_e$ and $\alpha = \frac{e^2}{\hbar c}$, and by measuring close lying levels great enhancement of

relative variation may be observed [13, 14, 15]. In particular, diatomic molecules that have a near cancellation between hyperfine structure and rotational intervals or between fine
structure and vibrational intervals are of interest in the context of such an enhancement (the transition frequency between two quasi-degenerated states are sensitive to the variation in fundamental constants; the enhancement is result of cancellation between the hyperfine and rotational intervals). A number of such molecules have been proposed, e.g. Cs₂ [16], CaH, MgH, CaH⁺ [17, 18], Cl⁺², IrC, HfF⁺, SiBr, LaS, LuO, and others [19].

55 Another interest of the rovibrational states of the LaS diatomic molecule is the possibilities of 56 precision tests of fundamental physics theories. Degeneracy between levels of different 57 symmetry happens frequently in diatomic molecules with unpaired electrons, such as LaS, 58 LaO, LuS, LuO, YbF and similar ions that have a cancellation between their hyperfine and 59 rotational interval [16].

Quantum computers based upon the principles of quantum superposition and entanglement are expected to provide more powerful computation ability than classical ones [20]. To realize quantum information processing, many schemes of quantum circuits have been proposed. Two of these schemes are i) The vibrational and rotational modes of diatomic molecule can be used to construct elementary quantum gates and to encode two qubits [21], and ii) the electric dipole moments of diatomic molecules, oriented along or against an

66 external electric field consists of the qubits [22].

A promising approach for realizing a quantum computer is based on using the vibrational
 states of molecules to represent qubits and tailored femtosecond laser pulses to implement
 quantum gate operations [23-25].

70 The important connection between energy relations of solids and molecules [26] and the 71 absence of electronic structure with rovibrational study and dipole moment calculation of 72 higher excited states of the molecule LaS stimulated us, in the present work, to investigate more extensive ab initio calculation of this molecule. Based on our previous theoretical 73 74 calculation [27-36], the potential energy curves and the spectroscopic constants of the low 75 lying 23 electronic states have been calculated where 19 new electronic states have been 76 studied here for the first time. The transition energy with respect to the minimum energy for 77 the ground state T_e, the equilibrium internuclear distance r_e , the harmonic frequency ω_{e} , the 78 rotational constant Be, and the permanent dipole moment curves have been obtained for the 79 considered electronic states of this molecule. Taking advantage of the electronic structure of the investigated electronic states of the LaS molecule and by using the canonical functions 80 approach [37-40], the eigenvalue E_v, the rotational constant B_v, and the abscissas of the 81 turning points r_{min} and r_{max} have been calculated for several vibrational levels of the 82 considered electronic states. These investigated results are expected to be useful for further 83 84 investigations of the LaS spectra and for the astronomers to search the presence of more

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transition metal sulfides in the atmosphere of cool stars, the possibilities for precision tests of fundamental physics theories, and the quantum information processing.

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88 2. Computational approach

89 2.1 Electronic state calculation

90 The potential energy curves of the lowest-lying electronic states of the LaS molecule have 91 been investigated via CASSCF method. Multireference CI calculations (single and double excitations with Davidson corrections) were performed. The sulfur atom is treated in all 92 93 electron schemes where the 16 electrons of sulfur atom are considered using for s, p, d, and 94 f functions of the ROOS basis set [41]. For the Lanthanum species the basis set ECP46MHF [42] has been used for the s, p, and d functions with effective core potential of 46 inner 95 electrons and 11 valance electrons. The potential energy calculations for the states $^{2s+1}\Lambda^{(+)}$ 96 of the molecule LaS have been carried using CAS-SCF method. Among the 27 electrons 22 97 98 electrons were frozen in subsequent calculations so that 5 electrons were explicitly treated 99 as valence electrons. The active space contains $2\sigma(\text{La: 5d}_0; \text{S: 3p}_0), 2\pi(\text{La: 5d}_{\pm 1}; \text{S: 3p}_{\pm 1}),$ 1δ (La: 5d₊₂) orbitals in the C_{2v} symmetry; this corresponds to 8 active molecular orbitals 100 distributed into irreducible representation a₁, b₁, a₂, b₂ in the following way: 3a₁, 2b₁, 2b₂, 1a₂, 101 102 noted [3,2,2,1]. This calculation has been performed via the computational chemistry 103 program MOLPRO [43] taking advantage of the graphical user interface GABEDIT [44].

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105 2.2 The vibration-rotation calculation

106 Within the Born-Oppenheimer approximation, the vibration rotation motion of a diatomic 107 molecule in a given electronic state is governed by the radial Schrödinger equation

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$$\left[\frac{d^2}{dr^2} + \frac{2\mu}{\hbar^2}(E_{\nu J} - U(r)) - \frac{\lambda}{r^2}\right] \Psi_{\nu J}(r) = 0$$
(1)

109 where r is the internuclear distance, v and J are respectively the vibrational and rotational 110 quantum numbers, $\lambda = J(J+1)$ E_{vJ} and Ψ_{vJ} are respectively the eigenvalue and the 111 eigenfunction of this equation. In the perturbation theory these functions can be expanded as

112
$$E_{\nu J} = \sum_{n=0}^{\infty} e_n \lambda^n$$
(2)

113
$$\Psi_{vJ}(r) = \sum_{n=0}^{\infty} \phi_n(r)\lambda^n$$
(3)

114 with $e_0 = E_v$, $e_1 = B_v$, $e_2 = D_v$, ϕ_0 is the pure vibrational wave function and ϕ_h its 115 rotational corrections. By replacing Eqs.(2) and (3) into Eq.(1) and since this equation is 116 satisfied for any value of λ , one can write [37-39]

117
$$\phi_0^{"}(r) + [e_0 - U(r)]\phi_0(r) = 0$$
 (4)

118
$$\phi_1''(r) + [e_0 - U(r)]\phi_1(r) = -[e_1 - R(r)]\phi_0(r)$$
(5-1)

119
$$\phi_{2}(r) + [e_{0} - U(r)]\phi_{2}(r) = -[e_{1} - R(r)]\phi_{1}(r) - e_{2}\phi_{0}(r)$$
120 (5-2)

121 $\phi_n''(r) + [e_0 - U(r)]\phi_n(r) = R(r)\phi_{n-1} - \sum_{m=1}^n e_m \phi_{n-m}(r)$ (5-n)

where $R(r)=1/r^2$, the first equation is the pure vibrational Schrödinger equation and the 122 remaining equations are called the rotational Schrödinger equations. One may project 123 124 Eqs.(7) onto ϕ_0 and find

125
$$<\phi_0 | \phi_0 > e_1 = <\phi_0 | \frac{1}{r^2} | \phi_0 >$$
 (6-1)

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$$<\phi_0 | \phi_0 > e_2 = <\phi_0 | \frac{1}{r^2} | \phi_1 > -e_1 < \phi_0 | \phi_1 >$$
 (6-2)

128
$$<\phi_0 | \phi_0 > e_n = <\phi_0 | \frac{1}{r^2} | \phi_{n-1} > -\sum_{m=1}^{n-1} e_m <\phi_{n-m} | \phi_0 >$$
 (6-n)

129 Once e_0 is calculated from Eq.(4), e_1 , e_2 , e_3 ... can be obtained by using alternatively Eqs.(5) 130 and (6). 131

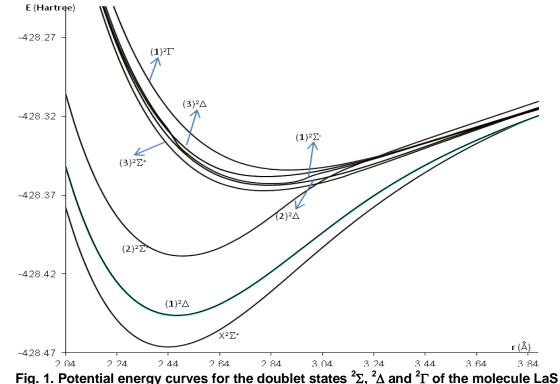
3. RESULTS AND DISCUSSION 132

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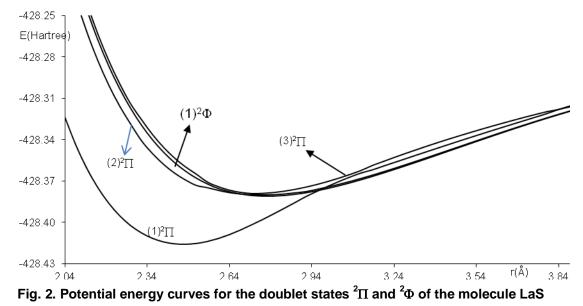
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The potential energy curves for 23 states in the representation $^{2s+1}\Lambda^{(+)}$ have been calculated 134

- for 47 internuclear distances in the range 2.00Å≤r≤3.44 Å. These PECs for the doublet and 135
- quartet states in the different symmetries are drawn respectively in Figs. (1-4). 136



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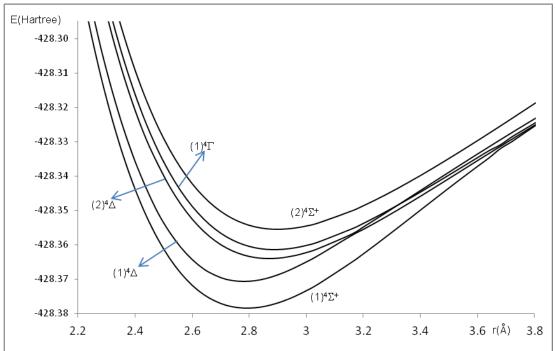
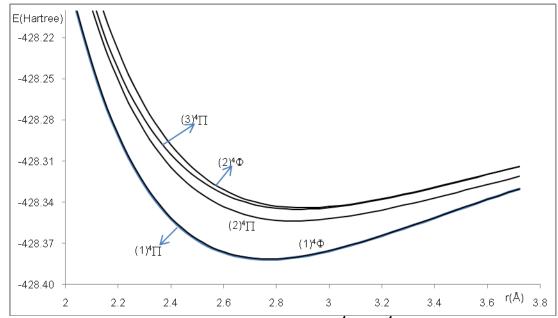


Fig. 3. Potential energy curves for the quartet states ${}^{4}\Sigma$, ${}^{4}\Delta$ and ${}^{4}\Gamma$ of the molecule LaS

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143 Fig. 4. Potential energy curves for the quartet states ${}^{4}\Pi$ and ${}^{4}\Phi$ of the molecule LaS

145 It is quite common to study a crossing or an avoided crossing, known as conical 146 intersections, between ground and excited state potential energy curves of molecules which 147 are important points in photochemistry. Such crossings or avoided crossings can dramatically alter the stability of molecules. In fact, the avoided crossing regions are likely to 148 be a leakage channels along which the molecules flow from the higher down to the lower 149 potential energy curves. If these crossings are overlooked then low barrier transitions can be 150 missed and an incorrect chemical picture will arise [45]. In the range of r considered, several 151 crossings have been detected in the potential energy curves for the excited electronic states 152 at short and large values of internuclear distance and for the highest and lowest states, the 153 154 positions of these crossings r_c are given in Table 1.

155

156	Table1. Positions of the crossings between the different electronic states of the
157	molecule LaS

state1	state2	Crossing between (n1)state1/(n2)state2	r _c (Å)
$X^2\Sigma$ +	$^{2}\Delta$	1/1	3.84
	$^{2}\Delta$	2/3	3.05
		2/4	3.24
		3/2	2.60
		3/4	2.06
	$^{2}\Sigma$	2/3	3.02
		2/4	3.20
$^{2}\Sigma$		3/4	2.46
2	² П	3/3	3.33
	$^{4}\Delta$	3/2	2.57
		3/1	3.66
		4/2	3.16
		4/1	3.01
	${}^{4}\Gamma$	3/1	3.64

		4/1	3.22
	$^{2}\Sigma$	1/2	3.25
		1/3	3.79
		1/4	3.41
	$^{2}\Delta$	1/2	3.72
		1/3	3.67
$^{2}\Gamma$		1/4	3.30
1	² П	1/3	3.70
	$^{4}\Delta$	1/2	3.45
	⁴ П	1/2	2.82
	11	1/2	2.35
	² П	2/3	3.43
	11	3/3	3.57
	$^{4}\Delta$	2/2	2.93
	Δ	2/1	3.56
$^{2}\Delta$		<u>-</u> /. 4/1	2.76
		3/2	2.95
		3/1	3.64
	${}^{4}\Gamma$	2/1	3.67
		4/1	3.04
	$^{2}\Phi$	1/1	2.99
		2/1	2.69
	$^{4}\Sigma$	3/1	3.11
	${}^{4}\Gamma$	3/1	3.65
² Π	$^{4}\Delta$	3/1	3.51
	$^{4}\Phi$	1/1	2.97
		2/1	2.63
	⁴ Π	1/1	2.97
		2/1	2.60
$^{4}\Delta$	${}^{4}\Gamma$	2/1	3.34
	$^{4}\Delta$	2/1	3.08
	$^{4}\Sigma$	3/2	2.43
⁴ Π	$^{4}\Delta$	2/1	2.31
	${}^{4}\Gamma$	2/1	2.57
		3/1	2.16

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The spectroscopic constants such as the vibrational harmonic constant ω_e , the internuclear distance at equilibrium r_e , the rotation constant B_e , and the electronic transition energy with respect to the ground state T_e have been calculated by fitting the energy values around the equilibrium position to a polynomial in terms of the internuclear distance, these values are displayed in Table 2 together with the available values in the literature either theoretical or experimental.

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166Table 2. Calculated spectroscopic constants for the lowest 23 electronic states of the
LaS molecule.

State	T _e (cm⁻¹)	ΔT _e /T _e %	^ω e (cm ⁻¹)	∆∞₀/∞₀ %	B _e (cm⁻¹)	ΔB _e /B _e %	r _e (Å)	∆r₀/r _e %
$X^{2}\Sigma^{+}$	0.00 ^a 0.00 ^b		434.48	4.67	0.10875		2.4395	
	0.00 ^c 0.00 ^d		456.70	4.67 1.04			2.37 ^d	2.83

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	0.00 ^{e1}		456.7				2.38 ^{e1}	2.42
	0.00 ^f 0.00 ^g		439 ^d 490 ^{e1}	12.78	0.117155 ^f 0.116976 ^g	7.72 7.03	2.23 ^{e2}	8.57
$(1)^2 \Delta$	4329.46		40.4.00		0.10568		2.4745	
(1) ² ∏	11043.47 ^a 10487.17 ^b	5.03	424.82 421.41 412.79	2.04	0.10547		2.4771	
(2) ² Σ	12696.59 ^a 13793.21 ^b 13790.17 ^c 13766.79 ^f	7.90 7.93 8.43	416.55 410.12 410.07	1.54 1.54	0.10546	5.3	2.4983	
(1) ⁴ ∏	13766.79 ⁹ 18587.35	8.43	286.03		0.11103 0.08449	5.3	2.7677	
$(1)^{4}\Phi$	18666.11		285.94		0.08437		2.7697	
$(1)^{2}\Phi$	18762.72		285.12		0.08376		2.7797	
(2) ² ∏	18932.00		251.61		0.08428		2.7706	
(3) ² Π (1) ⁴ Σ	19184.48 ^a 22694 ^b 20388.91	15.46	352.23 344 288.54	2.34	0.08642 0.08295		2.7366 2.7931	
$(1)^{2}\Sigma$	21791.09		272.97		0.08160		2.8145	
$(0) \perp$ $(1)^4 \Delta$	22116.21		296.97		0.08363		2.7819	
$(2)^2\Delta$	22509.87		265.13		0.08043		2.8364	
$(3)^2\Delta$	22564.00		271.96		0.08060		2.8336	
$(4)^2\Sigma$	22726.51		321.94		0.08056		2.8342	
$(2)^4 \Delta$	23588.60		248.21		0.07956		2.8530	
$(4)^2 \Delta$	23713.00		250.50		0.08114		2.824	
(1) ⁴ Γ	24128.83		280.75		0.07746		2.8901	
(1) ² Γ	24633.02		246.57		0.07652		2.9081	
(2) ⁴ ∏	24747.86		261.71		0.07830		2.8747	
(2) ⁴ Σ	25408.26		277.44		0.07701		2.8986	
(3) ⁴ ∏	26591.66		268.99		0.07762		2.8873	
(2) ⁴ Φ	26865.56		237.65		0.07688		2.9016	

a)Present work, b)Ref.[9], c)Ref.[8], d)Ref.[4], e1)Ref.[47 empirical method 1], e2)Ref.[47 empirical method 2], f)Ref.[11], g)Ref.[10]

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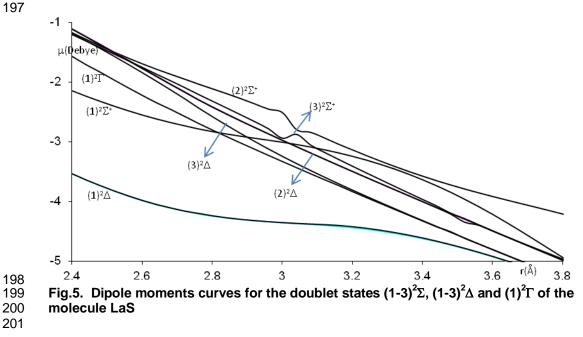
171 The comparison of our calculated values with the few fragmented results in the literature relative difference 172 shows good agreement for Te with а 5.03% Ref.[9]< $\Delta T_e/T_e$ <8.43% Ref.[10] for the states(1)² Π and (2)² Σ^+ . This agreement 173 becomes less for the state $(3)^2\Pi$ with $\Delta T_e/T_e = 15.46\%$. For the vibrational harmonic constant 174 ω_e , our calculated values are in good agreement for the published values for the studied 175 states in literature with relative difference 1.04%Ref.[4]<\Delta\u0e_ke_s4.67%Ref.[9] except the 176

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177 value given in Ref.[47] where the relative difference is 12.78%. The three published values in literature for each of B_e (for the states $X^2\Sigma^+$ and $(2)^2\Sigma$ and r_e (for the state $X^2\Sigma^+$) showed a 178 179 aood agreement with our calculated values with relative differences 180 5.3%Refs.[10,11]<∆Be/Be<7.72%Ref.[11] and

181 2.42%Refs.[$\frac{47}{(\text{method 1})} < \Delta r_e/r_e < 8.57\%$ Ref.[$\frac{47}{(\text{method 2})}$] respectively. The comparison for 182 the other calculated values is not possible since they are given here for the first time.

183 The permanent dipole moment operator is among the most reliably predicted physical 184 properties. DeMille [25] described the molecular qubits as permanent electric dipoles 185 oriented along |0> or against |1> in an external electric field. Bits are coupled by the electric 186 dipole-dipole interaction. Loading with ultracold molecules makes it possible to use a weak 187 trapping potential, which should allow long decoherence times for the system. The 188 expectation value of the permanent dipole moment is sensitive to the nature of the least energetic and most chemically relevant valence electrons [47-49]. Since the Hartree-Fock 189 190 dipole moment is usually large, and in order to obtain the best accuracy, multireference 191 configuration interaction (MRCI) wave functions were constructed using multi configuration 192 Self-consistent field (MCSCF) active space. The permanent dipole moments of the investigated electronic states have been analyzed for the 23 lowest electronic states of the 193 molecule LaS. The calculated values of the dipole moments (in Debye) for these electronic 194 states in terms of the internuclear distance r are plotted in figures (5-9). All these calculations 195 196 were performed by using the MOLPRO [40] program.



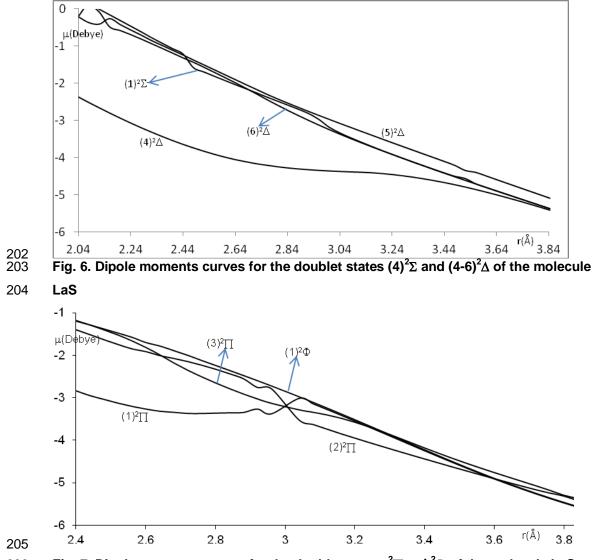


Fig. 7. Dipole moments curves for the doublet states ${}^{2}\Pi$ and ${}^{2}\Phi$ of the molecule LaS.

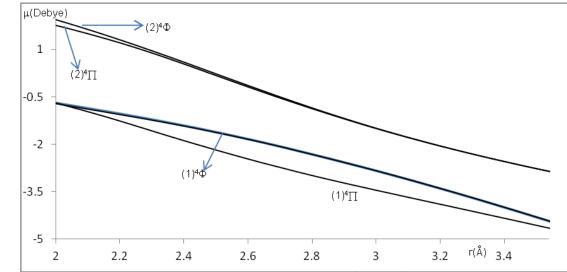
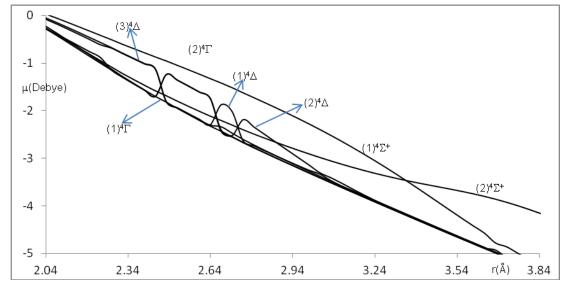




Fig. 8. Dipole moments curves for the quartet states ${}^{4}\Pi$ and ${}^{4}\Phi$ of the molecule LaS.





210 Fig. 9. Dipole moments graphs for the quartet states ${}^{4}\Sigma$, ${}^{4}\Gamma$, and ${}^{4}\Delta$ of the molecule LaS By using the canonical functions approach and the cubic spline interpolation between each two 211 consecutive points of the potential energy curves obtained from the ab initio calculation of the 212 LaS molecule, the eigenvalue E_v, the rotational constant B_v, and the abscissas of the turning 213 point r_{min} and r_{max} have been calculated up to the vibrational levels v=20 for the considered 214 electronic states. These values for the states $(1)^{2}\Delta$, $(1)^{2}\Pi$, $(1)^{4}\Pi$, $(2)^{2}\Sigma$ (as illustration) are given 215 in Table 3. No comparison of these values with those in literature since they are given here for 216 217 the first time.

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Table 3. Values of the eigenvalues E_v , the abscissas of the turning points r_{min} , r_{max} , and the rotational constant B_v for the different vibrational levels of the $(1)^2 \Delta$, $(1)^2 \Pi$, $(1)^4 \Pi$, $(2)^2 \Sigma$ of the LaS molecule.

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		(1) ² Δ					(1) ² ∏	
V	E _v (cm⁻¹)	r _{min} (Å)	r _{max} (Å)	B _v ×10 (cm-1)	Е _v (cm-1)	r _{min} (Å)	r _{max} (Å)	B _v ×10 (cm-1)
0	215.35	2.417	2.529	1.0583	210.89	2.419	2.531	1.0567
1	625.60	2.379	2.574	1.0549	620.12	2.391	2.575	1.0537
2	1036.24	2.354	2.606	1.0525	1029.19	2.356	2.607	1.0517
3	1445.29	2.335	2.632	1.0493	1436.73	2.336	2.634	1.0483
4	1853.59	2.318	2.656	1.0470	1842.22	2.319	2.657	1.0464
5	2261.01	2.304	2.677	1.0443	2246.44	2.304	2.679	1.0435
6	2667.70	2.290	2.697	1.0414	2647.88	2.290	2.699	1.0412
7	3072.03	2.279	2.716	1.0382	3046.32	2.278	2.719	1.0375
8	3474.14	2.268	2.735	1.0350	3441.73	2.267	2.738	1.0339
9	3874.15	2.258	2.752	1.0321	3833.24	2.257	2.756	1.0296
10	4272.53	2.248	2.764	1.0291	4222.04	2.248	2.774	1.0257
11	4669.38	2.239	2.785	1.0263	4607.97	2.239	2.791	1.0223
12	5064.72	2.230	2.801	1.0234	4992.02	2.231	2.808	1.0187
13	5458.56	2.223	2.816	1.0204	5373.49	2.223	2.824	1.0151
14	5850.64	2.215	2.832	1.0174	5751.86	2.215	2.841	1.0111
15	6240.93	2.208	2.846	1.0142	6127.25	2.208	2.857	1.0070
16	6629.41	2.201	2.861	1.0112	6499.81	2.873	2.873	1.0031
17					6868.81	2.195	2.888	0.9935
		(1) ⁴ ∏			(2) ² Σ			
v	E _v (cm⁻¹)	r _{min} (Å)	r _{max} (Å)	B _v × <mark>10</mark> ²	Ev.	r _{min}	r _{max}	B _v ×10
	(cm⁻¹)			(cm ⁻¹)	(cm ⁻¹)	(Å)	(Å)	(cm ⁻¹)
				. ,	· · /	()	()	、
0	142.50	2.702	2.836	8.4375	206.46	2.440	2.554	1.03768
0 1	142.50 429.32	2.702 2.657	2.836 2.890		. ,			
1				8.4375	206.46	2.440	2.554	1.03768
1 2	429.32	2.657	2.890	8.4375 8.4085	206.46 607.34	2.440 2.402	2.554 2.599	1.03768 1.0351
1	429.32 715.48	2.657 2.628	2.890 2.930	8.4375 8.4085 8.3781	206.46 607.34 1008.10	2.440 2.402 2.377	2.554 2.599 2.631	1.03768 1.0351 1.0321
1 2 3 4	429.32 715.48 999.44	2.657 2.628 2.606	2.890 2.930 2.962	8.4375 8.4085 8.3781 8.3489	206.46 607.34 1008.10 1412.61	2.440 2.402 2.377 2.357	2.554 2.599 2.631 2.657	1.03768 1.0351 1.0321 1.0302
1 2 3 4 5 6	429.32 715.48 999.44 1282.48	2.657 2.628 2.606 2.586	2.890 2.930 2.962 2.992	8.4375 8.4085 8.3781 8.3489 8.3151	206.46 607.34 1008.10 1412.61 1814.77	2.440 2.402 2.377 2.357 2.340	2.554 2.599 2.631 2.657 2.680	1.03768 1.0351 1.0321 1.0302 1.0288
1 2 3 4	429.32 715.48 999.44 1282.48 1564.12	2.657 2.628 2.606 2.586 2.570	2.890 2.930 2.962 2.992 3.019	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849	206.46 607.34 1008.10 1412.61 1814.77 2219.00	2.440 2.402 2.377 2.357 2.340 2.325	2.554 2.599 2.631 2.657 2.680 2.701	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259
1 2 3 4 5 6 7	429.32 715.48 999.44 1282.48 1564.12 1843.83	2.657 2.628 2.606 2.586 2.570 2.554 2.541	2.890 2.930 2.962 2.992 3.019 3.044 3.068	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.740	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233
1 2 3 4 5 6 7 8	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77	2.657 2.628 2.606 2.586 2.570 2.554	2.890 2.930 2.962 2.992 3.019 3.044	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67	2.440 2.402 2.377 2.357 2.340 2.325 2.312	2.554 2.599 2.631 2.657 2.680 2.701 2.721	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205
1 2 3 4 5 6 7 8 9	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.721 2.740 2.758	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160
1 2 3 4 5 6 7 8	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69	2.657 2.628 2.606 2.586 2.570 2.554 2.554 2.541 2.528	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.721 2.740 2.758 2.775	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185
1 2 3 4 5 6 7 8 9 10	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69 2945.67	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517 2.506	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113 3.135	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612 8.1302	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56 4213.97	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278 2.268	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.721 2.740 2.758 2.775 2.791	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160 1.0137
1 2 3 4 5 6 7 8 9 10 11	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69 2945.67 3216.99	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517 2.506 2.496	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113 3.135 3.135	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612 8.1302 8.0993	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56 4213.97 4610.15	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278 2.268 2.268 2.259	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.740 2.758 2.775 2.791 2.807	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160 1.0137 1.0113
1 2 3 4 5 6 7 8 9 10 11 12	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69 2945.67 3216.99 3486.67	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517 2.506 2.496 2.486	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113 3.135 3.135 3.155 3.175 3.195	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612 8.1302 8.0993 8.0685 8.0380	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56 4213.97 4610.15 5005.45	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278 2.268 2.259 2.251 2.244	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.740 2.758 2.775 2.791 2.807 2.822	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160 1.0137 1.0113 1.0088 1.0062
1 2 3 4 5 6 7 8 9 10 11 12 13	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69 2945.67 3216.99 3486.67 3754.70	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517 2.506 2.496 2.486 2.476	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113 3.135 3.155 3.155 3.155 3.175 3.195 3.215 3.234	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612 8.1302 8.0993 8.0685	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56 4213.97 4610.15 5005.45 5399.43	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278 2.268 2.259 2.251 2.244 2.235 2.227	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.740 2.758 2.775 2.791 2.807 2.822 2.834	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160 1.0137 1.0113 1.0088 1.0062 1.0035 1.0007
1 2 3 4 5 6 7 8 9 10 11 12 13 14	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69 2945.67 3216.99 3486.67 3754.70 4021.08	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517 2.506 2.496 2.486 2.476 2.468	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113 3.135 3.135 3.155 3.175 3.195 3.215	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612 8.1302 8.0993 8.0685 8.0380 8.0075	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56 4213.97 4610.15 5005.45 5399.43 5792.01	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278 2.268 2.259 2.251 2.244 2.235	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.740 2.758 2.775 2.791 2.807 2.822 2.834 2.851	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160 1.0137 1.0113 1.0088 1.0062 1.0035
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69 2945.67 3216.99 3486.67 3754.70 4021.08 4285.79	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517 2.506 2.496 2.496 2.486 2.476 2.468 2.460	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113 3.135 3.155 3.155 3.155 3.175 3.195 3.215 3.215	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612 8.1302 8.0993 8.0685 8.0380 8.0075 7.9770	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56 4213.97 4610.15 5005.45 5399.43 5792.01 6183.02	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278 2.268 2.259 2.251 2.244 2.235 2.227	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.740 2.758 2.775 2.791 2.807 2.822 2.834 2.851 2.866	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160 1.0137 1.0113 1.0088 1.0062 1.0035 1.0007
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69 2945.67 3216.99 3486.67 3754.70 4021.08 4285.79 4548.85 4810.28	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517 2.506 2.496 2.496 2.486 2.476 2.468 2.460 2.452 2.444	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113 3.135 3.155 3.155 3.175 3.195 3.215 3.215 3.234 3.252 3.271	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612 8.1302 8.0993 8.0685 8.0380 8.0075 7.9770 7.9436 7.9155	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56 4213.97 4610.15 5005.45 5399.43 5792.01 6183.02 6572.09 6959.08	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278 2.268 2.259 2.251 2.244 2.235 2.227 2.220 2.213	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.740 2.758 2.775 2.791 2.807 2.822 2.834 2.851 2.866 2.880 2.894	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160 1.0137 1.0113 1.0088 1.0062 1.0035 1.0007 0.9977 0.9945
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16	429.32 715.48 999.44 1282.48 1564.12 1843.83 2121.77 2398.05 2672.69 2945.67 3216.99 3486.67 3754.70 4021.08 4285.79 4548.85	2.657 2.628 2.606 2.586 2.570 2.554 2.541 2.528 2.517 2.506 2.496 2.496 2.476 2.468 2.460 2.452	2.890 2.930 2.962 2.992 3.019 3.044 3.068 3.091 3.113 3.135 3.155 3.155 3.175 3.195 3.215 3.234 3.234 3.252	8.4375 8.4085 8.3781 8.3489 8.3151 8.2849 8.2545 8.2234 8.1923 8.1612 8.1302 8.0993 8.0685 8.0380 8.0075 7.9770 7.9436	206.46 607.34 1008.10 1412.61 1814.77 2219.00 2620.21 3019.67 3148.47 3816.56 4213.97 4610.15 5005.45 5399.43 5792.01 6183.02 6572.09	2.440 2.402 2.377 2.357 2.340 2.325 2.312 2.300 2.289 2.278 2.268 2.259 2.251 2.244 2.235 2.227 2.220	2.554 2.599 2.631 2.657 2.680 2.701 2.721 2.740 2.758 2.775 2.791 2.807 2.822 2.834 2.851 2.866 2.880	1.03768 1.0351 1.0321 1.0302 1.0288 1.0259 1.0233 1.0205 1.0185 1.0160 1.0137 1.0113 1.0088 1.0088 1.0062 1.0035 1.0007 0.9977

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224 **4. Conclusion**

In the present work, the *ab initio* investigation for 23 low-lying electronic states of LaS molecule has been performed via CAS-SCF/MRCI method. The potential energy and permanent dipole moments curves have been determined along with the spectroscopic constants T_e , r_e , ω_e and the rotational constant B_e for the lowest-lying electronic states. Based on the canonical function approach a rovibrational calculation has been performed for the investigated electronic states. The comparison of our results with those available in the

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literature has demonstrated the good accuracy except the value of T_e for the state $(3)^2\Pi$. Nineteen electronic states have been investigated in the present work for the first time. As these results are expected to be useful for further investigations of the LaS spectra, quantum computation and the fundamental physics theories, extensive energy values versus internuclear distance r and the rovibrational energy levels of the ground and excited electronic states are available upon request.

237 238 **REFERENCES**

Freedmann H, Emge TJ, Brennan JG. (THF)₈Ln₈E₆(EPh)₁₂ Cluster Reactivity:
 Systematic Control of Ln, E, EPh, and Neutral Donor Ligands, Inorg. Chem. 1999; 38:
 4400-4404.

242 2. Freedman D, Sayan S, Emge TJ, Croft M, Brennan JG, Heterovalent Clusters: 243 $Ln_4Se(SePh)_8 (Ln_4 = Sm_4, Yb_4, Sm_2Yb_2, Nd_2Yb_2)$, J. Am. Chem. Soc. 1999; 121: 11713-244 11719.

3. Melman JH, Fitzgerald M, Freedman D, Emge TJ, Brennan JG, Chalcogen-Rich
Lanthanide Clusters from Lanthanide Halide Starting Materials: A New Approach to the
Low-Temperature Synthesis of LnS_x Solids from Molecular Precursors, J. Am. Chem.
Soc. 1999; 121: 10247-10248.

[4] Luo Yi, Xiaohong Wan, Yuki Ito, Seiichi Takami, Momoji Kubo, Akira Miyamoto, A
density functional theory calculation on lanthanide monosulfides, Chem. Phys. 2002;
282: 197206.

5. Harrison JF, Electronic Structure of Diatomic Molecules Composed of a First-Row Transition Metal and Main-Group Element (H-F), Chem. Review 2000; 100: 679-716.

6. Hinkel KH, Lambert DL, and Wing RF, Mon. Not. R. Astron. Soc. 1989; 238: 1365.

255 7. Lambert DL and Celgg RES, Mon. Not. R. Astron. Soc. 1980; 191: 367.

256 8. Macrano M and Barrow RF, Analysis of the transition B² Σ^+ -X² Σ^+ in gaseous LaS, J. 257 phys. B 1970; 3: L121-L122.

258 9. Winkel RJ Jr., Davis SP, and Abrams MC, Fourier transform spectroscopy of LaS in the 259 infrared, Appl. Opt. 1996; 35: 2874-2878.

10. Anderson N, Davis SP, Edvinsson G, and Winkel RJ Jr., Rotational Analysis of the LaS Spectrum in the Region 5000–15000 cm⁻¹, phys. Scr. 2001; 64: 134-139.

262 11. He SG, Tam WS, Leung JW-H, and Cheung AS-C, Laser spectroscopy of LaS:

263 Hyperfine structure in the $B^2\Sigma^+ - X^2\Sigma^+$ (0,0) band, J. Chem. Phys. 2002; 117: 5764-5769.

- 12. Hendrick L, Bethlem L, and Ubachs W, Testing the time-invariance of fundamental
- constants using microwave spectroscopy on cold diatomic radicals, The Royal Soc. of Chem.
 Farad. Discuss 2009; 142: 25-36.
- 267 13. Flambaum VV, Berengut JC, Space-time variation of coupling constants and
- fundamental masses, International journal of modern physics A 2009; 24: 3342 .

14. Chin C, Flambaum VV, and Kozlov MG, Ultracold molecules: new probes on the variation of fundamental constants, New J. Phys. 2009; 11: 055048.

- 271 15. Murphy MT, Flambaum VV, Muller S, and Henkel C, Strong limit on a variable proton-
- to-electron mass ratio from molecules in the distant universe, Science 2008; 320: 1611-1613.
- 273 16. DeMille D, Sainis S, Sage J, Bergeman T, Kotochigova S, and Tiesinga E, Enhanced
- 274 Sensitivity to Variation of m_e/m_p in Molecular Spectra, Phys. Rev. Lett. 2008; 100: 043202-275 043206.
- 276 17, Kajita M, Prospects of detecting m_e/m_p variance using vibrational transition frequencies
- 277 of ${}^{2}\Sigma$ -state molecules, Phys. Rev. A 2008; 77:012511-012522.
- 18. Kajita M and Moriwaki Y, Proposed detection of variation in m_p/m_e using a vibrational transition frequency of a CaH⁺ ion, J. Phys. B 2009; 42: 154022.

Corresponding author, Tel : +961 3 747617; Fax : +961 1 818 402 : E-mail address:

- 280 19. Flambaum VV and Kozlov MG, Enhanced Sensitivity to the Time Variation of the Fine-
- Structure Constant and m_p/m_e in Diatomic Molecules, Phys. Rev. Lett. 2007; 99: 150801-150805.
- 283 20. Tesch CM, Kurtz L, and de Vivie-Riedle R, Applying optimal control theory for
- elements of quantum computation in molecular systems, Chem. Phys. Lett. 2001; 343:633-641.
- 286 21. K. Mishima K, Tokumo K, and Yamashita K, Quantum computing using molecular
- electronic and vibrational states, Chem. Phys. 2008; 343: 61-75.
- 288 22. Zhao M and Babikov D, Phase control in the vibrational qubit, J. Chem. Phys. 2006;125:
 289 024105-024111.
- 23. Lee C. and Ostrovskaya EA, Quantum Computation with Diatomic Bits in Optical
 Lattices, arXiv:guant-ph/0502154v4 4 Nov 2005.
- 24. Babikov D, Accuracy of gates in a quantum computer based on vibrational eigenstates, J.
 Chem. Phys. 2004; 121: 7577-7586.
- 294 25. DeMille D, Quantum computation with trapped polar molecules, arXiv:quant-
- 295 ph/0109083v2 27 Oct 2001
- 296 26. Smith JR, Schlosser H, Leaf W, Ferrante J, Rose JH, Connection between energy
- relations of solids and molecules, Phys. Rev. A. 1989; 39: 514-517.
- 298 27. Abdul Al SN, Korek M, Allouche AR. Theoretical electronic structure of the lowest-lying 299 states of the YI molecule. Chem. Phys. 2005; 308: 1-6.
- 300 28. Korek M, Moghrabi YA, A. R. Allouche AR. Theoretical calculation of the excited states of
 301 the KCs molecule including the spin-orbit interaction. J. Chem. Phys. 2006; 124: 094309302 094310.
- 303 29. Korek M, Bleik S, Allouche AR. Theoretical calculation of the low laying electronic states 304 of the molecule NaCs with spin-orbit effect. J. Chem. Phys. 2007; 126: 124313-124322
- 305 30. Taher-Mansour F, Allouche AR, Korek M. Theoretical electronic structure of the lowestlying states of ScCl molecule below 22 500 cm⁻¹. J. Mol. Spectrosc. 2008; 248: 61-65.
- 307 31. Korek M, Kontar S, Taher-Mansour F, Allouche AR. Theoretical electronic structure of the molecule Scl. Int. J. Quant. Chem. 2009; 109: 236-242.
- 309 32. Korek M, Al-Shawa S, Younes GA. Theoretical calculation of the electronic structure of
- the molecule LiRb including the spin–orbit interaction. J. Mol. Struct. THEOCHEM 2009; 899:
 25-31.
- 312 33. Korek M and Abdul-Al SN. Rovibrational study and dipole moment calculation of the
 313 molecule YF with spin–orbit interaction. Chem. Phys. 2009; 355: 130-134.
- 314 34. Hamdan A and Korek M. Theoretical calculation of the low-lying sextet electronic states
 315 of CrF molecule. Chem. Phys. 2010; 369: 13-18.
- 316 35. Hamdan A and Korek M. Theoretical calculation of the low-lying quartet states of the CrF 317 molecule. Can. J. Chem. 2011; 89: 1304-13011.
- 318 36. Hamdan A and Korek M. Theoretical study with vibration–rotation and dipole moment 319 calculations of quartet states of the CrCl molecule. Int. J. Quant. Chem. 2011; 111: 2960-
- 320 2965.
- 321 37. Kobeissi H, Korek M, and Dagher M. On the computation of diatomic centrifugal
- distortion constants: Exact solutions for initial value problems. J. Mol. Spectrosc. 1989; 138:
 1-12.
- 324 38. Korek M and Kobeissi H. Highly accurate diatomic centrifugal distortion constants for 325 high orders and high levels. J. Comput. Chem. 1992; 13: 1103-1108.
- 326 39. Korek M, A one directional shooting method for the computation of diatomic centrifugal distortion constants. Comput. Phys. commun. 1999; 119: 169-178.
- 40. Tohme SN and korek M, Electronic Structure and Rovibrational Calculation
- 329 of the molecule RbYb, 2012, (*in Press*)

Corresponding author, Tel: +961 3 747617; Fax: +961 1 818 402: E-mail address:

- 330 41. MOLPRO Roos Basis set.
- 42. MOLPRO Basis Query, Pseudopotentials of the Stuttgart/Cologne group
 (Revision: July 09, 2009).
- 43. MOLPRO is a package of *ab initio* programs written by H.-J. Werner,
- P.J.Knowles, R. Lindh, F. R. Manby, M. Schütz, P. Celani, T. Korona, G. Rauhut,
 R. D. Amos, A. Bernhardsson, A. Berning, D. L. Cooper, M. J. O. Deegan, A. J.
- 336 Dobbyn, F. Eckert, C. Hampel, G. Hetzer, A. W. Lloyd, S. J. McNicholas, W.
- 337 Meyer, M. E. Mura, A. Nicklaß, P. Palmieri, R. Pitzer, U. Schumann, H. Stoll, A. J.
- 338 Stone, R. Tarroni, and T. Thorsteinsson.
- 44. Allouche AR, "Gabedit, A graphical user interface for computational chemistry softwares", J. Comput. Chem., 2011; 32:174-182.
- 45. Fedorov DG, Koseki S, Schmidt MW, Gordon MS, Spin-orbit coupling in molecules:
- Chemistry beyond the adiabatic approximation, Int. Rev. Phys. Chem. 2003; 22: 551-592.
- 46. Czack G, Hein H, Hinz I, Bergmann H, Kuhun P, Gmelin Handbook of Inorganic
- 345 Chemistry, C7, eighth edition, Spinger, New York, 1983.
- 346 47. Barenco A, Deutsch D, Ekert A, and Jozsa R, Conditional Quantum Dynamics and
- 347 Logic Gates, Phys. Rev. Lett. 1995; 74: 4083-4086.
- 48. Brennen GK, Caves CM, Jessen PS, and Deutsch IH, Quantum Logic Gates in Optical
- 349 Lattices, Phys. Rev. Lett. 1999; 82: 1060 -1063.
- 49. Platzman P and Dykman M, Quantum computing with electrons floating on liquid helium,
- 351 Science 1999; 284:1967-1969.