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2 **Original Research Article**

3 **Growth of CdS nanoparticles to fabricate**

4 **Schottky barrier**

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15 **ABSTRACT**

CdS nanoparticles have been grown by a simple cost effective chemical reduction method and a Schottky barrier of gold/ nano CdS is fabricated. The grown nanoparticles are structurally characterized by transmission electron microscopy and x ray diffraction. The optical properties of nano CdS is characterized by optical absorption, photoluminescence study. The band gap of the CdS nanoparticles is increased as compared to CdS bulk form. Capacitance–voltage and current–voltage characteristics of gold /nano CdS Schottky barrier junction have been studied. It is found that these characteristics are influenced by surface or interface traps. The values of barrier height, ideality factor, donor concentration and series resistance are obtained from the reverse bias capacitance–voltage measurements.

16 *Keywords: CdS nanoparticles; structural properties; optical properties; Au/n-CdS Schottky*

17 *barrier*

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20 **1. INTRODUCTION**

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22 Semiconductor nanoparticles are promising material in electrical and optoelectronic devices.

23 Properties of nanostructures such as structural, electrical, optical etc. are different from their

24 bulk form due to mainly quantum confinement effect and surface to volume ratio [1-

25 3].Cadmium Sulphide a group II - group VI semiconductor having characteristic band gap

26 2.42 eV in bulk form has been used in different optoelectronic devices e.g. Solar cell, LED,

27 Laser etc. [4-7]. CdS nanostructures based hetero junctions, Schottky barriers are important

28 for application in such devices [8-10].The Schottky barriers of CdS with different metals have

29 been studied by researchers [11-13].The electrical properties of Schottky devices are

30 affected by metal semiconductor interface or surface properties of semiconductors [14].

31 Proper modification of surface states of semiconducting nanoparticles is still a challenge for

32 researchers. In this work an effort has been made to grow CdS nanoparticles by a very cost

33 effective and controlled way. The effect of nanoparticles surface on the formation of barrier is

34 investigated to modify the Schottky device based on CdS nanoparticles. There are various

35 physical and chemical methods to prepare CdS nanoparticles [15-18]. We have followed a

36 simple chemical reduction method to grow CdS nanoparticles which is cost effective also

37 [19].

38 The structural and optical properties of synthesized CdS nanoparticles are characterized.

39 Schottky junction of gold (Au)/n-CdS has been fabricated. The electrical properties of Au/n-

CdS Schottky junction have been studied by current-voltage and capacitance – voltage measurements. The values of barrier height, ideality factor, and donor concentration are obtained by experiment results.

2. EXPERIMENTAL DETAILS

The CdS nanoparticles are grown by a chemical reduction method at room temperature. Cadmium chloride, sulphur powder and sodium borohydride are used to grow CdS nanoparticles. The structure of grown nanoparticles are characterized Transmission Electron Microscope JEOL JEM200 at 200 kV. Optical absorption of the grown nanoparticles is performed by Shimadzu-Pharmaspec-1700 visible and ultraviolet spectrophotometer. Photoluminescence spectra of sample are observed by Perkin Elmer spectrophotometer. The procedure to grow CdS nanoparticles, structural, optical characterization of as prepared CdS nanoparticles is described elsewhere [19, 20].

To fabricate Schottky junction a film of the CdS nanoparticles has been grown from the dispersed CdS nanoparticles on ITO coated glass. The pre-cleaned ITO coated glass substrate has been dipped in to the dispersed solution of CdS nanoparticles at least for 6 hrs. Uniformly thin film of CdS nanoparticles has been deposited on the glass substrate. Schottky junction is fabricated by evaporating gold (Au) dots of 2 mm diameter through a mask on CdS film. Fig.1 shows the schematic diagram of fabricated Schottky barrier

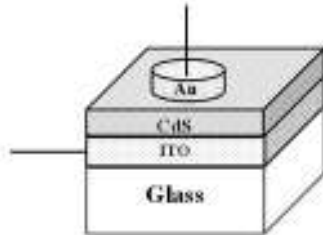


Fig.1 Schematic diagram of fabricated Au/nanoCdS Schottky device

Current-voltage and capacitance-voltage measurements of Au/n-CdS Schottky junction are performed using HP4284A LCR meter and Keithley electrometer.

3. RESULTS AND DISCUSSION

The TEM image and corresponding selected area electron diffraction (SAD) pattern of grown CdS nanoparticles is shown in Fig.2. The TEM image confirms that CdS nanoparticles are formed and agglomerated. The size of the as prepared nanoparticles is of the order of 11-14 nm. The SAD pattern displays the presence of diffraction rings which corresponds to the hexagonal wurtzite crystal phase of CdS.

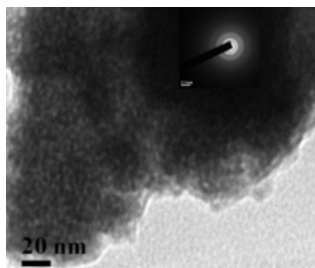


Fig. 2 TEM image and SAD pattern (inset) of as synthesized CdS nanoparticles

Fig. 3 shows the x ray diffraction (XRD) pattern of the as prepared sample. The XRD pattern shows that synthesized nano CdS sample has hexagonal wurtzite structure [20]. The prominent peaks shown in the XRD pattern are indexed with respective planes.

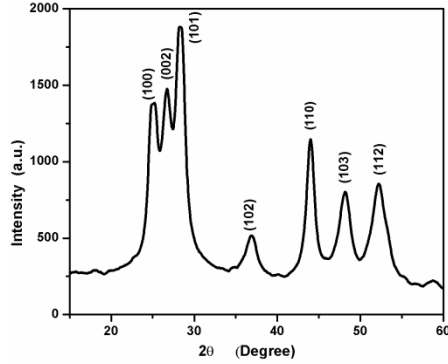


Fig. 3 The XRD pattern of the as grown CdS nanoparticles

The variation of optical absorbance of CdS nanoparticles with wavelength is shown in Fig. 4

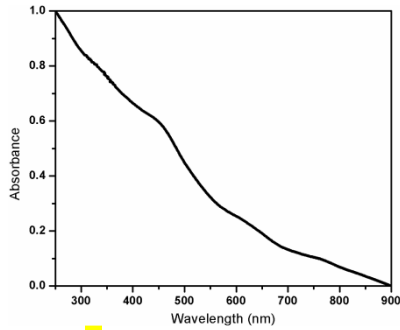


Fig. 4 The optical absorption spectrum of as prepared sample

The absorption spectrum is normalized. The band gap of the as-prepared nanoparticles is determined from the Tauc relation [21]

$$(\alpha h\nu)^2 = C (h\nu - E_g) \quad (1)$$

Where C is a constant. E_g is the band gap of the semiconductor material and α is the absorption coefficient. Band gap of the CdS nanoparticles is calculated from $(\alpha h\nu)^2$ vs. $h\nu$ plot which is given in figure 5. The linear part of the curve is extrapolated to energy ($h\nu$) axis to determine band gap. The band gap is found to be 2.97 eV.

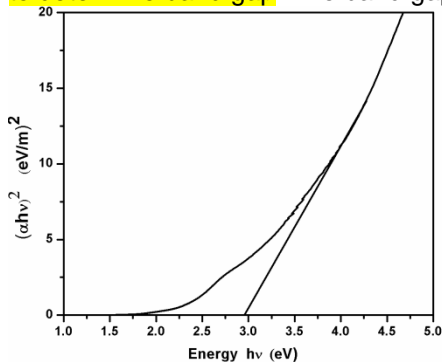


Fig. 5 The band gap determination curve for as prepared sample

The photoluminescence spectrum of as-prepared CdS sample is displayed in Fig. 6.

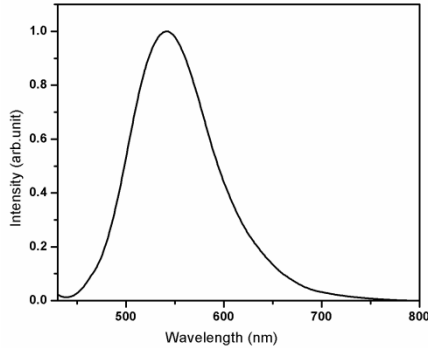


Fig.6 The photoluminescence spectra of as prepared sample

Photoluminescence intensity is normalized. Photoluminescence spectrum displays peak around 542 nm due to presence of surface states [22].

Capacitance(C) –Voltage (V) Measurement

The C-V measurement of Au/n-CdS Schottky junction with reverse and forward biasing voltages at temperature 303 K is shown in Fig. 7.

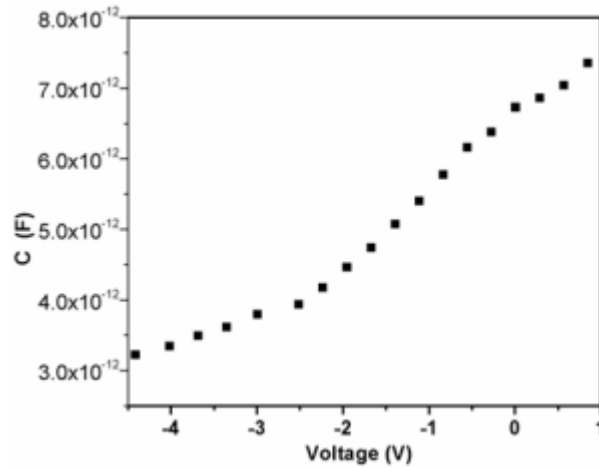


Fig. 7 The C-V characteristics of nano n-CdS /Au Schottky junction

The $1/C^2$ vs. V plot is given in figure 8. The carrier concentration, built-in-voltage is determined from the slope [23] and the intercept on the V axis of $1/C^2$ vs. V plot using the Mott-Schottky relation (2)

$$C^{-2} = \frac{2(V_b + V)}{q\epsilon\epsilon_0 A^2 N_d}$$

where N_d is the donor concentration, V_b is the built-in potential, q is the electronic charge, ϵ_0 is the permittivity of free space, ϵ is the dielectric constant of the semiconductor. W is the width of the depletion region. A is the area of the device. In Mott-Schottky relation it is assumed that surface or interface traps are absent, no interfacial layer is present between metal and semiconductor [14].

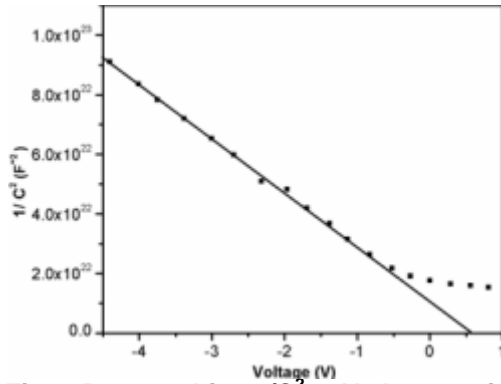


Fig. 8 Reverse bias $1/C^2$ vs V characteristics of Au/n-CdS Schottky barrier

The obtained values of N_d and V_b are given in Table 1. The value of barrier height ϕ_b is calculated by the following relation

$$\Phi_b = V_b + V_p \quad (3)$$

Where V_p is the potential difference between the Fermi level and the top of the valance band in CdS. V_p is calculated by knowing the donar concentration N_d and value of N_d is obtained from the following relation

$$V_p = KT \ln (N_c / N_d) \quad (4)$$

Where $N_c = 1.5 \times 10^{20} \text{ cm}^{-3}$ is the density of states in the conduction band for CdS [13]. The calculated barrier height value for the Au/n-CdS Schottky junction is given in table 1. It is seen from the result that C-V characteristics of Au/n-CdS Schottky junction is influenced by surface traps. [24,25].

Current (I)–voltage (V) characteristics

The I-V characteristics of the Au/n-CdS device under forward and reverse biasing conditions at 303K is shown in Fig. 9.

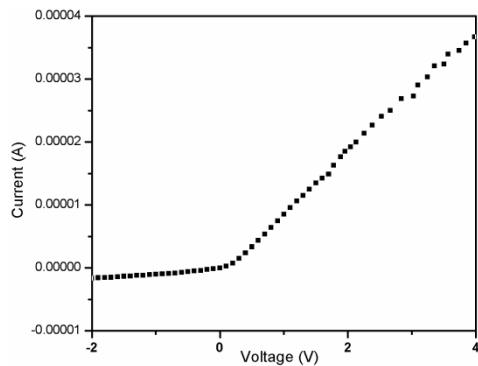


Fig. 9 The I-V characteristics of the Au/n-CdS device in forward and reverse biasing condition

The electron affinity of n-type CdS is 4.8 eV [26] while the work function of gold (Au) is about 5.25 eV [27]. So a Schottky barrier should be formed at the contact interfaces of Au/n-CdS.

146 For a metal semiconductor Schottky barrier diode assuming thermionic emission to be the
147 dominant transport mechanism the relationship between current and voltage is given by

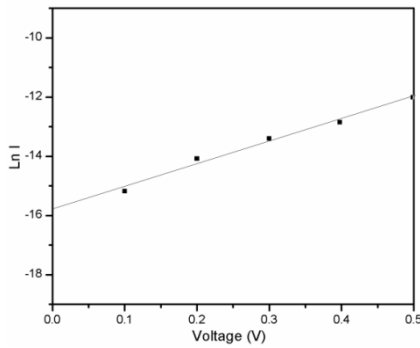
$$148 \quad I = I_s [\exp(qV/\eta kT) - 1] \quad (5)$$

149 Where I_s is saturation current, V is applied voltage, η is the ideality factor, k is the Boltzmann
150 constant, T is the absolute temperature in Kelvin. I_s is described as

$$151 \quad I_s = AA^* T^2 \exp(-q\Phi/kT) \quad (6)$$

152 Where A is the area of device, A^* is the modified Richardson constant and Φ is the effective
153 barrier height from metal to semiconductor.

154 The saturation current is determined from a plot of $\ln(I)$ vs voltage (V), where I_s is obtained
155 as the intercept of the linear region of the $\ln(I)$ vs V curve extrapolated to zero voltage.
156 Figure 10 shows the current voltage characteristics $\ln(I)$ vs V plot. The saturation current is
157 evaluated 1.42×10^{-7} A.



158

159 **Fig. 10 Current-voltage characteristics of the nano-CdS/Au Schottky barrier plotted as**
160 **$\ln(I)$ - V**

161 Taking into consideration of series resistance R_s the equation can be expressed as

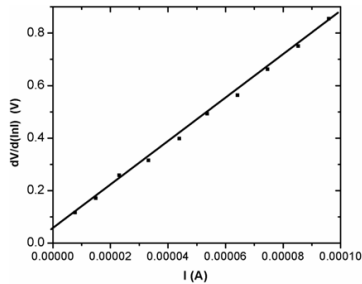
$$162 \quad I = I_s [\exp(q(V - IR_s)/\eta kT) - 1] \quad (7)$$

163 The equation can be differentiated as [28]

$$164 \quad dV/d(\ln I) = IR_s + \eta kT/q \quad (8)$$

165 The plot associated with Eq. (8) $dV/d(\ln I)$ vs I is given in Fig. 11.

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167

168 **Fig. 11 $dV/d(\ln I)$ versus I plot for Au/n-CdS device.**

169 The series resistance R_s is calculated from the slope of $dV/d(\ln I)$ versus I characteristic
 170 according to equation (7) [29]. The series resistance R_s here includes the contact resistance.
 171 While η is evaluated from the $dV/d(\ln I)$ axis intercept of the line fit shown in Fig. 9. The series
 172 resistance is found to be 8.27 k Ω . The ideality factor in the room temperature is listed in
 173 Table 1.

174 **Table 1 Different parameters of Au/n-CdS Schottky junction at temperature 303K**

V_b (V)	N_d (cm ⁻³)	Φ (eV)	η
0.56	5.41×10^{15}	.82	2.19

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 182 The obtained high series resistance of the Au/n-CdS Schottky device may be attributed to
 183 the high resistance of the starting CdS material or to the interfacial layer created between
 184 the metal and CdS [4]. The ideality factor is determined to be 2.19. Which is greater than
 185 typical value between 1 to 2 [30]. But the values of ideality factor greater than 2 is also
 186 possible [31]. Patel et al found ideality factor of Au/n CdS Schottky barrier 1.8, 6.0 [14].
 187 Ideality factor greater than 2 has been obtained with Schottky devices made of
 188 nanostructures. An oxide layer may be present between semiconductor and metal [32]. The
 189 high value of η may be due to large recombination within the interfacial layer [15, 33] which
 190 exists mainly in the semiconductor side.

191 **4. CONCLUSION**

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 193 We have synthesized CdS nanoparticles by a cost effective chemical method. The structural
 194 and optical characterizations of the synthesized CdS nanoparticles have been done. We
 195 have also fabricated Au/n- CdS Schottky junction with the grown CdS nanoparticles. The C-
 196 V and I-V characteristics of the Au/n-CdS Schottky junction have been studied. The values
 197 of built in potential, saturation current, barrier height, ideality factor, series resistance, the
 198 density of interface states have been calculated. It is found that the I-V and C-V
 199 characteristics of the Au/n-CdS Schottky junction are influenced by the surface states or
 200 interface traps.

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