ABSTRACT

MHD free convection, heat and mass transfer flow over a rotating inclined permeable plate with the influence of magnetic field, thermal radiation and chemical reaction of various order has been investigated numerically. The governing boundary-layer equations are formulated and transformed into a set of similarity equations with the help of similarity variables derived by lie group transformation. The governing equations are solved numerically using the Nactsheim-Swigert Shooting iteration technique together with the Runge-Kutta six order iteration schemes. The simulation results are presented graphically to illustrate influence of magnetic parameter (M), porosity parameter (γ) , rotational parameter (R'), Grashof number (G_r) , modified Grashof number (G_m) , thermal conductivity parameter (T_c) , Prandtl number (P_r) , radiation parameter (R), heat source parameter (Q), Eckert number (E_c) , Schmidt number (S_c) , reaction parameter (λ) and order of chemical reaction (n) on the all fluid velocity components, temperature and concentration distribution as well as Skin-friction coefficient, Nusselt and Sherwood number at the plate.

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Keywords: MHD; Inclined permeable plate; Thermal radiation; Chemical reaction;

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NOMENCLATURE

15		
16	B_0	Constant magnetic flux density
17	С	Constant depends on the properties of the fluid
18	С	Concentration of the fluid
19	C_{p}	Specific heat at constant pressure
20	D_m	Mass diffusivity
21	f'	Dimensionless primary velocity
22	g	Acceleration due to gravity
23	g_0	Dimensionless secondary velocity
24	k	Thermal conductivity
25	$k_{\scriptscriptstyle \infty}$	Undisturbed thermal conductivity
26	k_{0}	Reaction rate

27	7	K	Permeability of the porous medium	
28	8	n	Order of chemical reaction	
29	9	P	Pressure distribution in the boundary layer	
30	0	q_r	Radiative heat flux in the <i>y</i> direction	
31	1	Q_T	Heat generation	
32	2	Q_0	Heat source	
33	3	t	Time	
34	4	Τ	Fluid temperature	
35	5	U	Uniform velocity	
36	6	U, V	Velocity components along x and y axes respectively	
37	7	x'	Dimensionless axial distance along x axis	
38	3	Dimensionless parameters		
39	9	E_c	Eckert number	
40	0	R'	Rotational parameter	
41	1	G_r	Grashof number	
42	2	G_m	Modified Grashof number	
43	3	М	Magnetic parameter	
44	4	P_r	Prandtl number	
45	5	Q	Heat source parameter	
46	6	R	Radiation parameter	
47	7	S_c	Schmidt number	
48	8	T_c	Thermal conductivity parameter	
49	9	γ	Permeability of the porous medium	
50	0	λ	Reaction parameter	
51	1			
52	2	Greek Symbo	ols	
53	3	v	Kinematic viscosity of the fluid	
54	4	μ	Dynamic viscosity of the fluid	
55	5	σ	Electrical conductivity	
56	6	σ_0	Constant electrical conductivity	
57	7	σ_{s}	Stefan-Boltzmann constant	
58	3	ρ	Density of the fluid	

59	α	Thermal diffusivity
60	$\alpha_1 - \alpha_6$	Arbitrary real number
61	β	Inclination angle
62	$oldsymbol{eta}_T$	Thermal expansion coefficient
63	$oldsymbol{eta}_C$	Concentration expansion coefficient
64	κ^*	Mean absorption coefficient
65	ε	Parameter of the group
66	Ψ	Stream function
67	η	Similarity variable
68	θ	Dimensionless temperature
69	φ	Dimensionless concentration
70	Ω	Angular velocity of the plate
71	Subscripts	
72	W	Condition of the wall
73	∞	Condition of the free steam

1. INTRODUCTION

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Coupled heat and mass transfer problems in the presence of chemical reactions are of importance in many processes and have, therefore, received considerable amount of attention of researchers in recent years. Chemical reactions can occur in processes such as drying, distribution of temperature and moisture over agricultural fields and groves of fruit trees, damage of crops due to freezing, evaporation at the surface of a water body, energy transfer in a wet cooling tower and flow in a desert cooler. Chemical reactions are classified as either homogeneous or heterogeneous processes. A homogeneous reaction is one that occurs uniformly throughout a given phase. On the other hand, a heterogeneous reaction takes a restricted area or within the boundary of a phase. Analysis of the transport processes and their interaction with chemical reactions is quite difficult and closely related to fluid dynamics. Chemical reaction effects on heat and mass transfer has been analyzed by many researchers over various geometries with various boundary conditions in porous and nonporous media. Symmetry groups or simply symmetries are invariant transformations that do not alter the structural form of the equation under investigation which is described by Bluman and Kumei [1]. MHD boundary layer equations for power law fluids with variable electric conductivity is studied by Helmy [2]. In the case of a scaling group of transformations, the group-invariant solutions are nothing but the well known similarity solutions which is studied by Pakdemirli and Yurusoy [3]. Symmetry groups and similarity solutions for free convective boundary-layer problem was studied by Kalpakides and Balassas [4]. Makinde [5] investigated the effect of free convection flow with thermal radiation and mass transfer past moving vertical porous plate. Seddeek and Salem [6] investigated the Laminar mixed convection adjacent to vertical continuously stretching sheet with variable viscosity and variable thermal diffusivity. Ibrahim. Elaiw and Bakr [7] studied the effect of the chemical reaction and radiation absorption on the unsteady MHD free convection flow past a semi infinite vertical permeable moving plate with heat source and suction. El-Kabeir, El-Hakiem and Rashad [8] studied Lie group analysis of unsteady MHD three dimensional dimensional by natural convection from an inclined stretching surface saturated porous medium. Rajeswari, Jothiram and Nelson [9] studied the effect of chemical reaction, heat and mass transfer on nonlinear MHD boundary layer flow through a vertical porous surface in the presence of suction. Chandrakala [10] investigated chemical reaction effects on MHD flow past an impussively started semi-infinite vertical plate. Joneidi, Domairry and Babaelahi [11] studied analytical treatment of MHD free convective flow and mass transfer over a stretching sheet with chemical reaction. Muhaimin, Kandasamy and Hashim [12] studied the effect of chemical reaction, heat and mass transfer on nonlinear boundary layer past a porous shrinking sheet in the presence of suction. Rahman and Salahuddin [13] studied hydromagnetic heat and mass transfer flow over an inclined heated surface with variable viscosity and electric conductivity. As per standard text and works of previous researchers, the radiative flow of an electrically conducting fluid and heat and mass transfer situation arises in many practical applications such as in electrical power generation, astrophysical flows, solar power technology, space vehicle re-entry, nuclear reactors.

The objective of this study is to present a similarity analysis of boundary layer flow past a rotating inclined permeable plate with the influence of magnetic field, thermal radiation, thermal conductivity and chemical reaction of various orders.

2. MATHEMATICAL MODEL OF THE FLOW AND GOVERNING EQUATIONS

Steady two dimensional MHD heat and mass transfer flow with chemical reaction and radiation over an inclined permeable plate y=0 in a rotating system under the influence of transversely applied magnetic field is considered. The x-axis is taken in the upward direction and y-axis is normal to it. Again the plate is inclined at an angle β with the x-axis. The flow takes place at $y \geq 0$, where y is the coordinate measured normal to the x-axis. Initially we consider the plate as well as the fluid is at rest with the same velocity $U\left(=U_{\infty}\right)$, temperature $T\left(=T_{\infty}\right)$ and concentration $C\left(=C_{\infty}\right)$. Also it is assumed that the fluid and plate is at rest after that the whole system is allowed to rotate with a constant angular velocity $R=\left(0,-\Omega,0\right)$ about the y-axis and then the temperature and species concentration of the plate are raised to $T_{w}\left(>T_{\infty}\right)$ and $C_{w}\left(>C_{\infty}\right)$ respectively, which are thereafter maintained constant, where T_{w} and C_{w} is the temperature and concentration respectively at wall and T_{∞} and C_{∞} is the temperature and concentration from the plate.

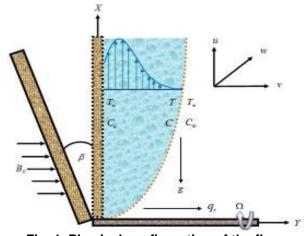


Fig. 1. Physical configuration of the flow

- 138 The electrical conductivity is assumed to vary with the velocity of the fluid and have the form
- 139 [2],
- 140 $\sigma = \sigma_0 u$, σ_0 is the constant electrical conductivity.
- 141 The applied magnetic field strength is considered, as follows [13]
- $142 B(x) = \frac{B_0}{\sqrt{x}}$
- 143 The temperature dependent thermal conductivity is assumed to vary linearly, as follows [6]
- 144 $k(T) = k_{\infty} \left[1 + c(T T_{\infty}) \right]$
- 145 Where k_{∞} is the undisturbed thermal conductivity and c is the constant depending on the
- 146 properties of the fluid.
- 147 The governing equations for the continuity, momentum, energy and concentration in laminar
- 148 MHD incompressible boundary-layer flow is presented follows

$$149 \qquad \frac{\partial u}{\partial x} + \frac{\partial v}{\partial y} = 0 \tag{1}$$

150
$$u\frac{\partial u}{\partial x} + v\frac{\partial u}{\partial y} = v\frac{\partial^2 u}{\partial y^2} + 2\Omega w - \frac{v}{K}u - \frac{\sigma_0 B_0^2 u^2}{\rho x} + g\beta_T (T - T_\infty)\cos\beta + g\beta_C (C - C_\infty)\cos\beta$$
 (2)

151
$$u\frac{\partial w}{\partial x} + v\frac{\partial w}{\partial y} = v\frac{\partial^2 w}{\partial y^2} - 2\Omega u - \frac{v}{K}w - \frac{\sigma_0 B_0^2 u w}{\rho x}$$
 (3)

152
$$u \frac{\partial T}{\partial x} + v \frac{\partial T}{\partial y} = \frac{1}{\rho C_n} \frac{\partial}{\partial y} \left[k \left(T \right) \frac{\partial T}{\partial y} \right] + \frac{Q_0 \left(T - T_{\infty} \right)}{\rho C_n} - \frac{\alpha}{k_{\infty}} \left(\frac{\partial q_r}{\partial y} \right) + \frac{\upsilon}{C_n} \left(\frac{\partial u}{\partial y} \right)^2$$
 (4)

153
$$u\frac{\partial C}{\partial x} + v\frac{\partial C}{\partial y} = D_m \frac{\partial^2 C}{\partial y^2} - k_0 \left(C - C_{\infty}\right)^n$$
 (5)

and the boundary conditions for the model is

- where, U is the uniform velocity, β is the inclination angle of the plate with x-axis, C_p is the
- specific heat at constant pressure, k(T) is the temperature dependent thermal conductivity,
- 158 Q_0 is the heat source, D_m is the mass diffusivity, k_0 is the reaction rate, $k_0 > 0$ for destructive
- reaction, $k_0 = 0$ for no reaction and $k_0 < 0$ for generative reaction, n (integer) is the order of
- 160 chemical reaction, q_r is the chemical reaction parameter, T_w and C_w is the temperature and
- 161 concentration respectively at wall and T_{∞} and C_{∞} is the temperature and concentration
- respectively far away from the plate.

2.1 METHOD OF SOLUTION

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166 Introducing the following dimensionless variables

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$$x' = \frac{xU}{v}, y' = \frac{yU}{v}, u' = \frac{u}{U}, v' = \frac{v}{U}, w' = \frac{w}{U}, \theta = \frac{T - T_{\infty}}{T_{w} - T_{\infty}} \text{ and } \varphi = \frac{C - C_{\infty}}{C_{w} - C_{\infty}}$$

the following equations are obtained,

169
$$u = U u', v = U v', w = U w', T = T_{\infty} + (T_{w} - T_{\infty})\theta$$
 and $C = C_{\infty} + (C_{w} - C_{\infty})\varphi$ (7)

Now, by using equation (7), the equations (1), (2), (3), (4) and (5) are transformed to

171
$$\frac{\partial u'}{\partial x'} + \frac{\partial v'}{\partial y'} = 0$$
 (8)

172
$$u'\frac{\partial u'}{\partial x'} + v'\frac{\partial u'}{\partial y'} = \frac{\partial^2 u'}{\partial y'^2} + 2R'w' - \gamma u' - \frac{Mu'^2}{x'} + G_r\theta\cos\beta + G_m\varphi\cos\beta$$
 (9)

173
$$u'\frac{\partial w'}{\partial x'} + v'\frac{\partial w'}{\partial y'} = \frac{\partial^2 w'}{\partial y'^2} - 2R'u' - \gamma w' - \frac{Mu'w'}{x'}$$
 (10)

174
$$u'\frac{\partial\theta}{\partial x'} + v'\frac{\partial\theta}{\partial y'} - \frac{1}{P_r} \left[(1 + T_c \theta + R) \frac{\partial^2\theta}{\partial y'^2} + T_c \left(\frac{\partial\theta}{\partial y'} \right)^2 \right] - Q\theta - E_c \left(\frac{\partial u}{\partial y} \right)^2 = 0$$
 (11)

175
$$u'\frac{\partial\varphi}{\partial x'} + v'\frac{\partial\varphi}{\partial y'} - \frac{1}{S_0}\frac{\partial^2\varphi}{\partial y'^2} + \lambda\varphi^n = 0$$
 (12)

using equation (7), the boundary condition (6) becomes

177
$$u' = 1, v' = 0, w' = 0, \theta = 1, \varphi = 1 \text{ at } y' = 0$$

$$u' \to 0, w' \to 0, \theta \to 0, \varphi \to 0 \text{ as } y' \to \infty$$
(13)

178 where

179
$$R' = \frac{\Omega v}{U^{2}}, \gamma = \frac{v^{2}}{KU^{2}}, M = \frac{\sigma_{0}B_{0}^{2}}{\rho}, G_{r} = \frac{g\beta_{r}(T_{w} - T_{\infty})v}{U^{3}}, G_{m} = \frac{g\beta_{c}(C_{w} - C_{\infty})v}{U^{3}}, T_{c} = c(T_{w} - T_{\infty}), T_{c} = c(T_{w}$$

180
$$R = \frac{16\sigma_S T_{\infty}^3}{3\kappa^* k_{\infty}}, P_r = \frac{v}{\alpha}, Q = \frac{Q_0 v}{\rho C_n U^2}, E_c = \frac{U^2}{C_n (T_w - T_{\infty})}, S_c = \frac{v}{D_m} \text{ and } \lambda = \frac{k_0 (C_w - C_{\infty})^{n-1} v}{U^2}$$

- 181 In order to deal with the problem, we introduce the stream function ψ (since the flow is
- 182 incompressible) defined by

183
$$u' = \frac{\partial \psi}{\partial v'}, v' = -\frac{\partial \psi}{\partial x'}$$
 (14)

- 184 The mathematical significance of using equation (14) is that the continuity equation (8) is
- 185 satisfied automatically.
- by equation (14), equations (9), (10), (11) and (12) transformed as follows,

187
$$\frac{\partial \psi}{\partial y'} \frac{\partial^2 \psi}{\partial x' \partial y'} - \frac{\partial \psi}{\partial x'} \frac{\partial^2 \psi}{\partial y'^2} - \frac{\partial^3 \psi}{\partial y'^3} - 2R'w' + \gamma \frac{\partial \psi}{\partial y'} + \frac{M}{x'} \left(\frac{\partial \psi}{\partial y'}\right)^2 - G_r \theta \cos \beta - G_m \phi \cos \beta = 0$$
 (15)

188
$$\frac{\partial \psi}{\partial y'} \frac{\partial w'}{\partial x'} - \frac{\partial \psi}{\partial x'} \frac{\partial w'}{\partial y'} - \frac{\partial^2 w'}{\partial y'^2} + 2R' \frac{\partial \psi}{\partial y'} + \gamma w' + \frac{M}{x'} \frac{\partial \psi}{\partial y'} w' = 0$$
 (16)

$$189 \qquad \frac{\partial \psi}{\partial y'} \frac{\partial \theta}{\partial x'} - \frac{\partial \psi}{\partial x'} \frac{\partial \theta}{\partial y'} - \frac{1}{P_r} \left[(1 + T_c \theta + R) \frac{\partial^2 \theta}{\partial y'^2} + T_c \left(\frac{\partial \theta}{\partial y'} \right)^2 \right] - Q \theta - E_c \left(\frac{\partial^2 \psi}{\partial y'^2} \right)^2 = 0$$

$$(17)$$

190
$$\frac{\partial \psi}{\partial y'} \frac{\partial \varphi}{\partial x'} - \frac{\partial \psi}{\partial x'} \frac{\partial \varphi}{\partial y'} - \frac{1}{S_c} \frac{\partial^2 \varphi}{\partial y'^2} + \lambda \varphi^n = 0$$
 (18)

and the boundary conditions (13) become,

192
$$\frac{\partial \psi}{\partial y'} = 1, \frac{\partial \psi}{\partial x'} = 0, w' = 0, \theta = 1, \varphi = 1 \text{ at } y' = 0$$

$$\frac{\partial \psi}{\partial y'} \to 0, w' \to 0, \theta \to 0, \varphi \to 0 \quad \text{as } y' \to \infty$$
(19)

- 193 Finding the similarity solution of the equations (15) to (18) is equivalent to determining the
- invariant solutions of these equations under a particular continuous one parameter group.
- 195 Introducing the simplified form of Lie-group transformations [8] namely, the scaling group of
- 196 transformations

197
$$G_1: x^* = x'e^{\mathcal{E}\alpha_1}, y^* = y'e^{\mathcal{E}\alpha_2}, \psi^* = \psi e^{\mathcal{E}\alpha_3}, w^* = w'e^{\mathcal{E}\alpha_4}, \theta^* = \theta e^{\mathcal{E}\alpha_5} \text{ and } \phi^* = \phi e^{\mathcal{E}\alpha_6}$$
 (20)

- 198 Here, $\varepsilon(\neq 0)$ is the parameter of the group and $\alpha's$ are arbitrary real numbers whose
- 199 interrelationship will be determined by our analysis. Equations (20) may be considered as a
- 200 point transformation which transforms the coordinates $(x', y', \psi, w', \theta, \varphi)$ to the coordinates
- 201 $(x^*, y^*, \psi^*, w^*, \theta^*, \varphi^*)$.
- The system will remain invariant under the group transformation G_1 , so the following
- relations among the exponents are obtained from equations (15) to (18),

$$\alpha_{1} + 2\alpha_{2} - 2\alpha_{3} = 3\alpha_{2} - \alpha_{3} = -\alpha_{4} = \alpha_{2} - \alpha_{3} = -\alpha_{5} = -\alpha_{6}$$

$$\alpha_{1} + \alpha_{2} - \alpha_{3} - \alpha_{4} = 2\alpha_{2} - \alpha_{4} = \alpha_{2} - \alpha_{3} = -\alpha_{4}$$

$$\alpha_{1} + \alpha_{2} - \alpha_{3} - \alpha_{5} = 2\alpha_{2} - \alpha_{5} = 2\alpha_{2} - 2\alpha_{5} = 4\alpha_{2} - 2\alpha_{3}$$

$$\alpha_{1} + \alpha_{2} - \alpha_{3} - \alpha_{6} = 2\alpha_{2} - \alpha_{6} = -n\alpha_{6}$$
(21)

205 Again, the following relations are obtained from the boundary conditions (19),

$$\begin{array}{ll}
\alpha_2 = \alpha_3 \\
\alpha_5 = \alpha_6 = 0
\end{array} \tag{22}$$

- 207 Solving the system of linear equations (21) and (22), the following relationship are obtained,
- 208 $\alpha_1 = 2\alpha_2 = 2\alpha_3, \ \alpha_4 = \alpha_5 = \alpha_6 = 0$
- 209 by using the above relation the equation (20) reduces to the following group of
- 210 transformation

211
$$x^* = x'e^{2\mathcal{E}\alpha_2}, y^* = y'e^{\mathcal{E}\alpha_2}, \psi^* = \psi e^{\mathcal{E}\alpha_2}, w^* = w', \theta^* = \theta, \phi^* = \phi$$
 (23)

- 212 expanding equation (23) by Taylor's method in powers of ε and keeping terms up to the
- 213 order ε , we have
- 214 $x^* x' = 2\varepsilon x' \alpha_2, y^* y' = \varepsilon y' \alpha_2, \psi^* \psi = \varepsilon \psi \alpha_2, w^* w' = 0, \theta^* \theta = 0, \varphi^* \varphi = 0$
- 215 In terms of differentials

216
$$\frac{dx'}{2\alpha_2 x'} = \frac{dy'}{\alpha_2 y'} = \frac{d\psi}{\alpha_2 \psi} = \frac{dw'}{0} = \frac{d\theta}{0} = \frac{d\varphi}{0}$$
 (24)

217 Solving the equation (24) the following similarity variables are introduced,

218
$$\eta = \frac{y'}{\sqrt{x'}}, \psi = \sqrt{x'} f(\eta), w' = g_0(\eta), \theta = \theta(\eta) \text{ and } \varphi = \varphi(\eta)$$

219 By using the above mentioned variables, equations (15), (16), (17) and (18) becomes

220
$$f''' + \frac{1}{2}ff'' - Mf'^{2} + 2R'g_{0} - \gamma f' + G_{r}\theta\cos\beta + G_{m}\phi\cos\beta = 0$$
 (25)

221
$$g_0'' + \frac{1}{2}fg_0' - 2Rf' - \gamma g_0 - Mf'g_0 = 0$$
 (26)

222
$$\frac{1}{P_c} (1 + T_c \theta + R) \theta'' + \frac{1}{P_c} T_c \theta'^2 + \frac{1}{2} f \theta' + Q \theta + E_c f''^2 = 0$$
 (27)

223
$$\frac{1}{S} \varphi'' + \frac{1}{2} f \varphi' - \lambda \varphi^n = 0$$
 (28)

224 The corresponding boundary conditions (19) become

225
$$\begin{cases} f' = 1, f = 0, g_0 = 0, \theta = 1, \varphi = 1 \text{ at } \eta = 0 \\ f' \to 0, g_0 \to 0, \theta \to 0, \varphi \to 0 \text{ as } \eta \to \infty \end{cases}$$
 (29)

where primes denote differentiation with respect to η only and the parameters are defined as

227
$$M = \frac{\sigma_0 B_0^2}{\rho}$$
 is the magnetic parameter,

228
$$\gamma = \frac{v^2 x'}{KU^2}$$
 is the porosity parameter

229
$$R' = \frac{\Omega v x'}{U^2}$$
 is the rotational parameter

230
$$G_r = \frac{g\beta_T (T_w - T_\infty) v x'}{U^3}$$
 is the Grashof number

231
$$G_m = \frac{g\beta_c (C_w - C_\infty) vx'}{U^3}$$
 is the modified Grashof number

232
$$T_c = c \left(T_w - T_{\infty} \right)$$
 is the thermal conductivity parameter

233
$$P_r = \frac{v}{\alpha}$$
 is the Prandtl number

234
$$R = \frac{16\sigma_s T_{\infty}^3}{3\kappa^* k_{\infty}}$$
 is the radiation parameter

235
$$Q = \frac{Q_0 v}{\rho C_p U^2}$$
 is the heat source parameter

236
$$E_c = \frac{U^2}{C_n (T_w - T_\infty)}$$
 is Eckert number

237
$$S_c = \frac{v}{D_m}$$
 is the Schmidt number

238
$$\lambda = \frac{k_0 \left(C_w - C_\infty \right)^{n-1} v}{U^2}$$
 is the reaction parameter

239 and n (integer) is the order of chemical reaction

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2.2 SKIN-FRICTION COEFFICIENTS, NUSSELT AND SHERWOOD NUMBER

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The physical quantities of the skin-friction coefficients, the reduced Nusselt number and reduced Sherwood number are calculated respectively by the following equations,

245
$$C_f(R_e)^{\frac{1}{2}} = -f''(0)$$
 (30)

246
$$C_{g_0}(R_e)^{\frac{1}{2}} = -g'_0(0)$$
 (31)

$$247 N_u \left(R_e \right)^{-\frac{1}{2}} = -\theta'(0) (32)$$

248
$$S_h(R_e)^{-\frac{1}{2}} = -\varphi'(0)$$
 (33)

249 where, $R_e = \frac{Ux'}{v}$ is the Reynolds number.

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3. RESULTS AND DISCUSSION

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In this paper heat and mass transfer problem associated with laminar flow past an inclined plate of a rotating system has been studied. In order to investigated the physical representation of the problem, the numerical values of primary velocity, secondary velocity, temperature and species concentration from equations (25), (26), (27) and (28) with the boundary layer have been computed for different parameters as the magnetic parameter

(M), the rotational parameter (R'), the porosity parameter (γ) , the Grashof number (G_r) , the modified Grashof number (G_m) , the radiation parameter (R), the Prandtl number (P_r) , the Eckert number (E_c) , the thermal conductivity parameter (T_c) , the heat source parameter (Q), the Schmidt number (S_c) , the reaction parameter (λ) , the inclination angle (β) and the order of chemical reaction (R) respectively.

It is observed from the Figs. 2a and 2b that as the magnetic parameter increases, the primary and secondary velocities are decreases and increases respectively. Figs. 3a-3d represent that with the increase of rotational parameter, primary velocity decreases but secondary velocity, temperature and concentration increases.

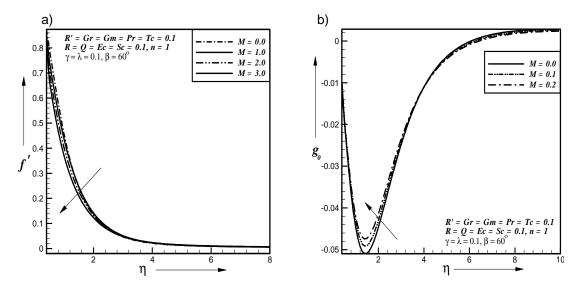
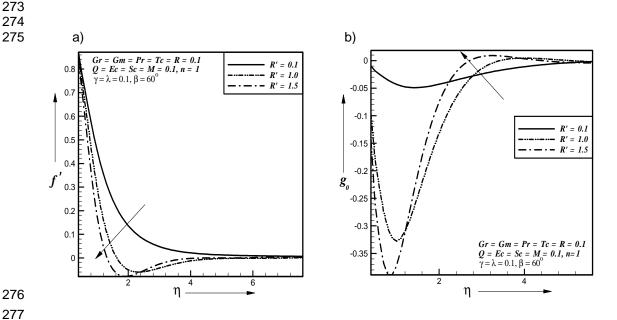


Fig. 2. Effect of magnetic parameter on a) primary velocity b) secondary velocity profiles





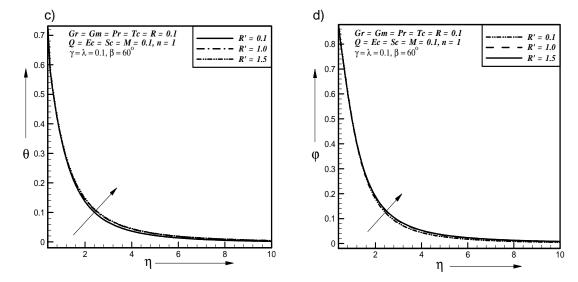
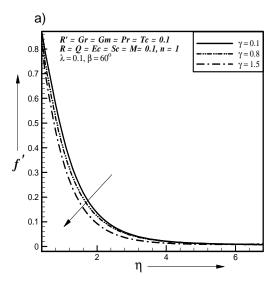
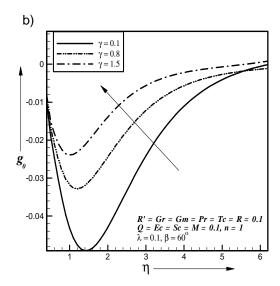


Fig. 3. Effect of rotational parameter on a) primary velocity b) secondary velocity c) temperature d) concentration profiles

From Figs. 4a, 4b, 4c and 4d we found that with the increase of porosity parameter, primary velocity decreases but the secondary velocity, temperature and concentration is increases respectively. Figs. 5a and 5b we observed that with the increase of inclination angle primary and secondary velocity profiles decreases and increases respectively.

It is observed from Figs. 6a, 6b and 6c that with the increase of Grashof number, primary velocity profile increases but secondary velocity and temperature profile decreases respectively. Figs. 7a, 7b and 7c show that with the increase of modified Grashof number, primary velocity increases but secondary velocity and concentration decreases. It is observed from Figs. 8a and 8b that with the increase of Prandtl number, primary velocity and temperature increases and decreases respectively.





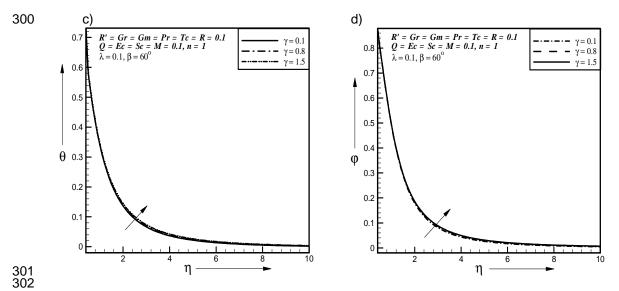


Fig. 4. Effect of porosity parameter on a) primary velocity b) secondary velocity c) temperature d) concentration profiles

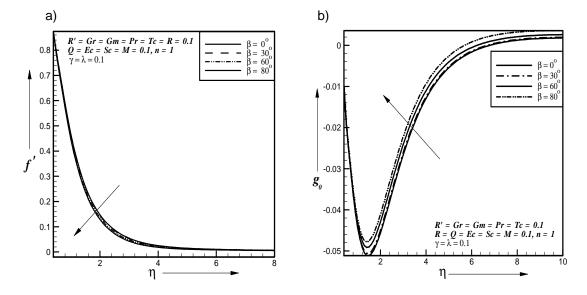


Fig. 5. Effect of inclination angle on a) primary velocity b) secondary velocity profiles

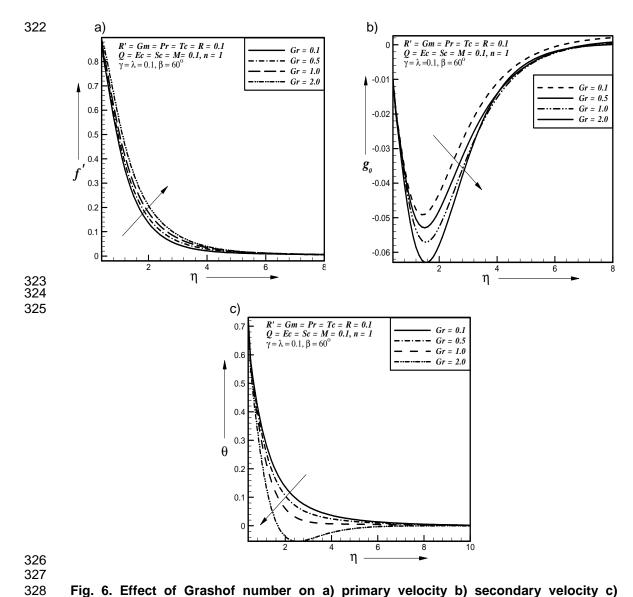


Fig. 6. Effect of Grashof number on a) primary velocity b) secondary velocity c) temperature profiles

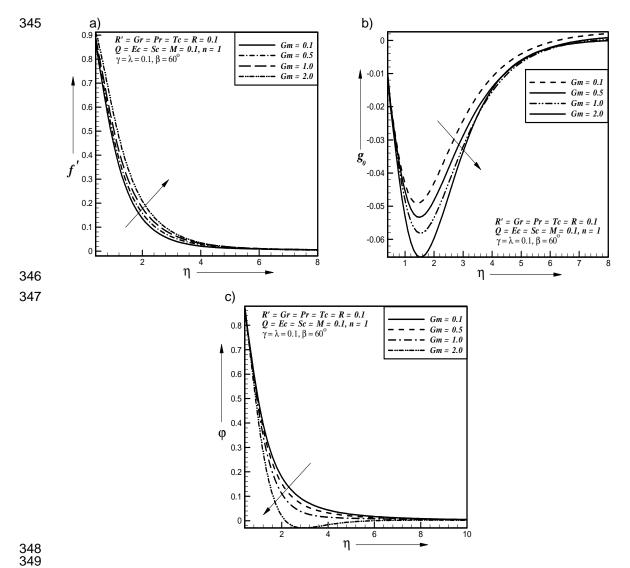


Fig. 7. Effect of modified Grashof number on a) primary velocity b) secondary velocity c) concentration profiles

Fig. 9a displays typical profiles for primary velocity (f') for different values of Eckert number. It is observed that the primary velocity is increases with the increase of Eckert number, where other parameters have the value $M = Gr = Gm = \gamma = Pr = Tc = R = Q = R' = Sc = 0.1$, $\lambda = 0.1$, $\beta = 60^{\circ}$, n = 1.

Fig. 9b displays typical profiles for temperature (θ) for different values of Thermal conductivity parameter. It is observed that the temperature is increases with the increase of Thermal conductivity parameter, where other parameters have the value $M = Gr = Gm = \gamma = Pr = 0.1$, $Ec = R = O = R' = Sc = \lambda = 0.1$, $\beta = 60^{\circ}$, n = 1.

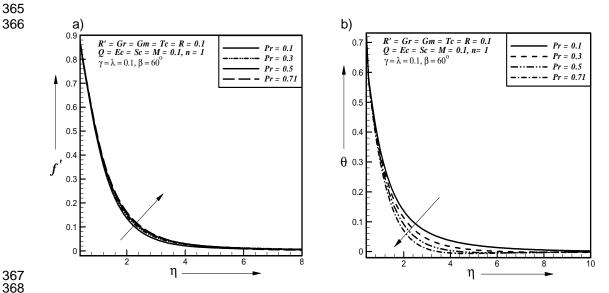


Fig. 8. Effect of Prandtl number on a) primary velocity b) temperature profiles

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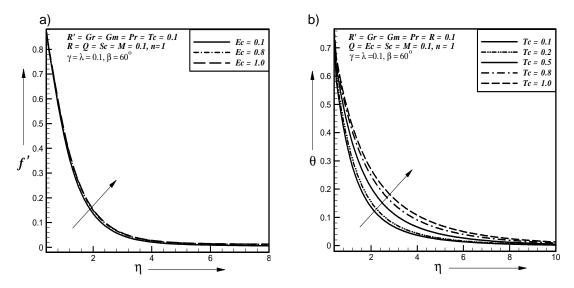


Fig. 9. Effect of a) Eckert number on primary velocity profiles b) thermal conductivity parameter on temperature profiles

Fig. 10a represents typical profiles for concentration (φ) for different values of Schmidt number Sc. It is observed that the concentration is decreases with the increase of Schmidt number. Fig. 10b represents no reaction $(\lambda = 0.0)$ and destructive reaction $(\lambda > 0.0)$ where we observed that the concentration is decreases with the increase of reaction parameter.



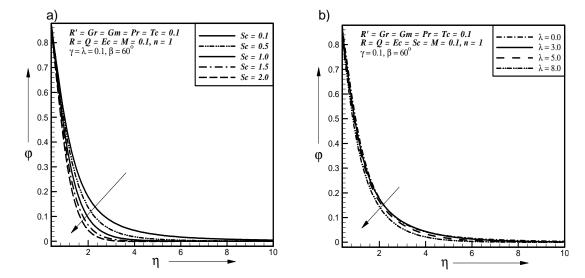


Fig. 10. Effect of a) Schmidt number on concentration profiles b) reaction parameter on concentration profiles

For the physical interest of the problem, the dimensionless skin-friction coefficient (-f'') and $(-g_0')$, the dimensionless heat transfer rate $(-\theta')$ at the plate and the dimensionless mass transfer rate $(-\varphi')$ at the plate are plotted against Heat source parameter (Q) and illustrated in Figs. 11-19.

Figs. 11a 1nd 11b represent the primary shear stress (-f'') and secondary shear stress $(-g_0')$ which are plotted against heat source parameter (Q) for different values of magnetic parameter. It is observed that the primary shear stress is decreases and secondary shear stress is increases with the increase of magnetic parameter, where other parameters have the value $R' = Gr = Gm = \gamma = Pr = Ec = Q = Tc = R = Sc = \lambda = 0.1, \beta = 60^0, n = 1$.

Figs. 12a and 12b represent the primary shear stress (-f'') and secondary shear stress $(-g_0')$ which are plotted versus heat source parameter (Q) for different values of rotational parameter. It is observed that the primary shear stress is decreases and secondary shear stress is increases with the increase of rotational parameter, where other parameters have the value.



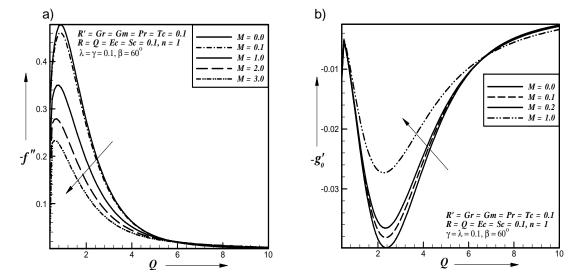


Fig. 11. Effect of magnetic parameter on a) primary shear stress b) secondary shear stress



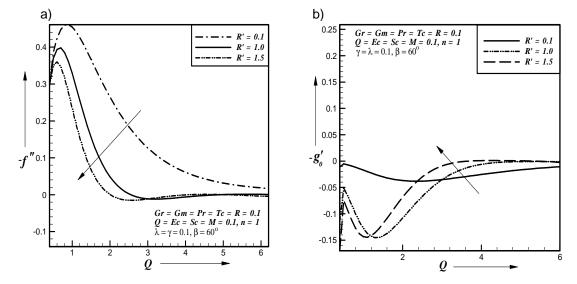


Fig. 12. Effect of rotational parameter on a) primary shear stress b) secondary shear stress

Figs. 13a and 13b represent the primary shear stress (-f'') and secondary shear stress $(-g_0')$ which are plotted versus heat source parameter (Q) for different values of porosity parameter. It is observed that the primary shear stress is decreases and secondary shear stress is increases with the increase of porosity parameter, where other parameters have the value $M = Gr = Gm = R' = Pr = Ec = Q = Tc = R = Sc = \lambda = 0.1, \beta = 60^0, n = 1$.



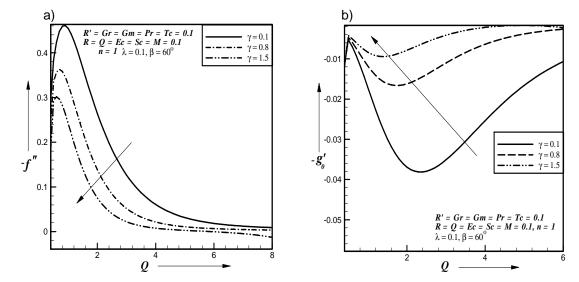


Fig. 13. Effect of porosity parameter on a) primary b) secondary shear stress

Fig. 14a represents the primary shear stress (-f'') which is plotted versus heat source parameter (Q) for different values of Grashof number. It is observed that the primary shear stress is increases with the increase of Grashof number, where other parameters have the value $M = \gamma = Gm = R' = Pr = Ec = Q = Tc = R = Sc = \lambda = 0.1, \beta = 60^{\circ}, n = 1$.

Fig. 14b represents the primary shear stress (-f'') which is plotted versus heat source parameter (Q) for different values of modified Grashof number. It is observed that the primary shear stress is increases with the increase of modified Grashof number, where other parameters have the value $M = \gamma = Gr = R' = Pr = Ec = Q = Tc = R = Sc = \lambda = 0.1, \beta = 60^{\circ}, n = 1.$

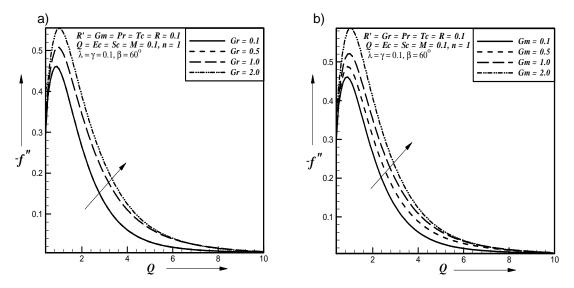


Fig. 14. Effect of a) Grashof number b) modified Grashof on primary shear stress

Fig. 15a represents the primary shear stress (-f'') which is plotted versus heat source parameter (Q) for different values of inclination angle. It is observed that the primary shear stress is decreases with the increase of inclination angle, where other parameters have the value $M = \gamma = Gm = Gr = R' = Pr = Ec = Q = Tc = R = Sc = \lambda = 0.1, n = 1$.

Fig. 15b represents the dimensionless heat transfer rate $(-\theta')$ which is plotted versus Heat source parameter (Q) for different values of thermal conductivity parameter. It is observed that the heat transfer rate is increases with the increase of thermal conductivity parameter, where other parameters have the value $M = \gamma = Gm = Gr = R' = Pr = Ec = Q = R = Sc = 0.1$,

 $\lambda = 0.1, \beta = 60^{\circ}, n = 1$.

Fig. 16a represents the dimensionless heat transfer rate $(-\theta')$ which is plotted versus heat source parameter (Q) for different values of Prandtl number. It is observes that the heat transfer rate is decreased with the increase of Prandtl number, where other parameters have the value $M = \gamma = Gm = Gr = R' = Tc = Ec = Q = R = Sc = \lambda = 0.1, \beta = 60^{0}, n = 1.$

Fig. 16b represents the dimensionless heat transfer rate $(-\theta')$ which is plotted versus Heat source parameter (Q) for different values of heat source parameter. It is observed that the heat transfer rate is increases with the increase of heat source parameter, where other parameters have the value $M = \gamma = Gm = Gr = R' = Tc = Ec = Pr = R = Sc = \lambda = 0.1, \beta = 60^{\circ}, n = 1$.

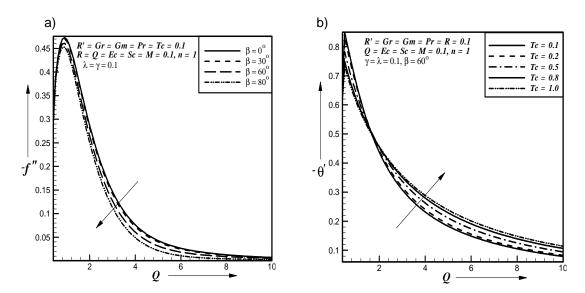


Fig. 15. Effect of a) inclination angle on primary shear stress b) thermal conductivity parameter on heat transfer rate



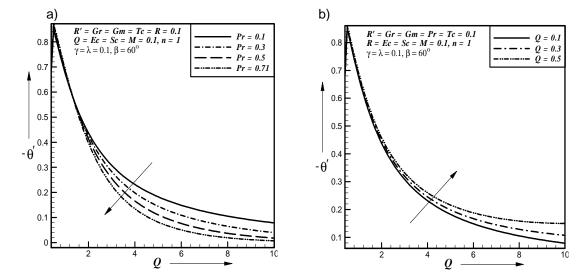


Fig. 16. Effect of a) Prandtl number b) heat source parameter on heat transfer rate

Fig. 17a and 17b, It is observed that the heat transfer rate increases with the increase of Eckert number and radiation parameter. Fig. 18a and Fig. 18b, show that mass transfer rate is decreases with the increase of Schmidt number and reaction parameter. Fig. 19 represents that the mass transfer rate increases with the increase of order of chemical reaction parameter.

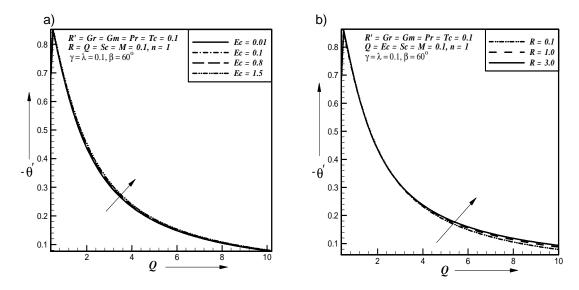


Fig. 17. Effect of a) Eckert number b) radiation parameter on heat transfer rate



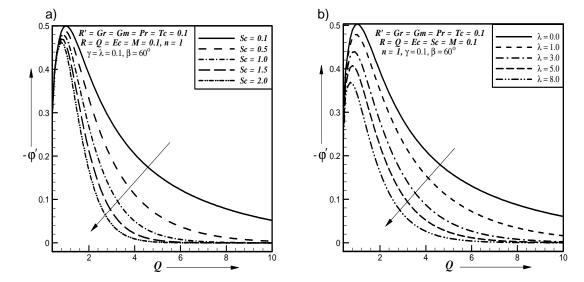


Fig. 18. Effect of a) Schmidt number b) reaction parameter on mass transfer rate

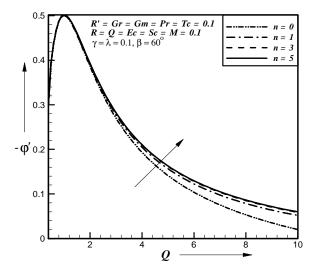


Fig. 19. Effect of order of chemical reaction on mass transfer rate

4. CONCLUSION

The primary velocity profiles are decreases with the increase of magnetic parameter but reverse effect is found for the secondary velocity profiles. Also the primary shear stress decreases due to increase of magnetic parameter where as the reverse effect is found for

The important findings of the investigation from graphical representation are listed below:

secondary shear stress.

The primary velocity profiles and primary shear stress are decreases due to increase of rotational parameter where as the reverse effect is found for the secondary velocity profiles

- and secondary shear stress. Also the temperature and concentration boundary layer thickness are increases due to increase of rotational parameter.
- The primary velocity profiles and primary shear stress are decreases due to increase of permeability of the porous medium where as the reverse effect is found for the secondary velocity profiles and secondary shear stress. Also the temperature and concentration
- 540 boundary layer thickness are increases due to increase of permeability of the porous 541 medium.
- The primary velocity profiles and primary shear stress are decreases due to increase of inclination angle where as the reverse effect is found for the secondary velocity profiles.
- 544 The primary velocity profiles and primary shear stress are increases due to increase of
- Grashof number where as the reverse effect is found for the secondary velocity profiles. Also
- the temperature boundary layer thickness is decreases due to increase of Grashof number.
- The primary velocity profiles and primary shear stress are increases due to increase of modified Grashof number where as the reverse effect is found for the secondary velocity
- 549 profiles. Also the concentration boundary layer thickness is decreases due to increase of modified Grashof number.
- 551 The primary velocity profiles are increases due to increase of Prandtl number. The thermal
- boundary layer thickness as well as the heat transfer rate at the plate is decreases as the
- 553 Prandtl number increases.
- The heat transfer rate at the plate as well as the primary velocity is increases due to increase of Eckert number.
- The temperature boundary layer thickness as well as the heat transfer rate at the plate is increases due to increase of thermal conductivity parameter.
- The heat transfer rate at the plate is increases due to increase of heat source parameter.
- The heat transfer rate at the plate is increases due to increase of radiation parameter.
- The concentration boundary layer thickness as well as the mass transfer rate at the plate is decreases due to increase of Schmidt number.
- The concentration boundary layer thickness as well as the mass transfer rate at the plate is decreases due to no reaction and destructive reaction.
- The mass transfer rate at the plate is increases due to increase of order of chemical reaction.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

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